

1 **Footprints of air pollution and changing environment on the sustainability**
2 **of built infrastructure**

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7 **Abstract**

8 Over 150 research articles relating three multi-disciplinary topics (air pollution, climate
9 change and civil engineering structures) are reviewed to examine the footprints of air
10 pollution and changing environment on the sustainability of building and transport
11 structures (referred as built infrastructure). The aim of this review is to synthesize the
12 existing knowledge on this topic, highlight recent advances in our understanding and
13 discuss research priorities. The article begins with the background information on
14 sources and emission trends of global warming (CO₂, CH₄, N₂O, CFCs, SF₆) and
15 corrosive (SO₂, O₃, NO_x) gases and their role in deterioration of building materials (e.g.
16 steel, stone, concrete, brick and wood) exposed in outdoor environments. Further
17 section covers the impacts of climate- and pollution-derived chemical pathways,
18 generally represented by dose-response functions (DRFs), and changing environmental
19 conditions on built infrastructure. The article concludes with the discussions on the
20 topic areas covered and research challenges. A comprehensive inventory of DRFs is
21 compiled. The case study carried out for analysing the inter-comparability of various
22 DRFs on four different materials (carbon steel, limestone, zinc and copper) produced
23 comparable results. Results of another case study revealed that future projected changes
24 in temperature and/or relative humidity are expected to have a modest effect on the
25 material deterioration rate whereas changes in precipitation were found to show a more
26 dominant impact. Evidences suggest that both changing and extreme environmental
27 conditions are expected to affect the integrity of built infrastructure both in terms of
28 direct structural damage and indirect losses of transport network functionality. Unlike
29 stone and metals, substantially limited information is available on the deterioration of
30 brick, concrete and wooden structures. Further research is warranted to develop more
31 robust and theoretical DRFs for generalising their application, accurately mapping
32 corrosion losses in an area, and costing risk of corrosion damage.

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34
35 **Keywords:** Air Pollutants; Built infrastructure; Dose-response functions; Climate
36 change; Green house and corrosive gases; Historic buildings; Transport infrastructure

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1. Introduction

Continuous anthropogenic emissions of greenhouse gases (GHG) into the atmosphere have raised the issue of changing climate. This is likely to alter the meteorology and result in important changes such as rise in global temperature, precipitation and sea level, alterations in ground water levels and soil conditions, and increased frequencies of extreme climate events (IPCC, 2007; Isaksen et al., 2009). Consequently, material environment (including buildings and transport infrastructure) is also likely to be affected. In the last decade or so, a large number of studies have studied the impact of climate change on the following key areas: (i) water and its related ecosystems such as storage reservoirs, waterways and irrigation channels, reticulated sewage systems, trunk sewers, treatment plants, storm water drains, flooding, lakes and fisheries (Aaheim et al., 1999; McIlgorm et al., 2010; Vincent and Gene, 2009), (ii) frequency and intensity of rainfall (Ekström et al., 2005), (iii) coastal and river flooding (Booij, 2005; Nicholls, 2004), (iv) power generation and transmission, gas and oil extraction, refining and distribution networks (Söderholm and Pettersson, 2008), (v) settlements and ice glaciers (Nayar, 2009), (vi) air quality and public health (Athanasiadou et al., 2010; Ebi and Burton, 2008; Haines et al., 2006; Vardoulakis and Heaviside, 2012), (vii) transport structures such as roads, railway lines, tunnels, bridges, earthworks, airports, ports, jetties, piers and sea wall protectors (Koetse and Rietveld, 2009; Larsen et al., 2008; TRB, 2008), and (viii) buildings such as historic, residential, commercial, industrial, storage, community and public space facilities (Brimblecombe and Grossi, 2007; Coley and Kershaw, 2010; Karaca, 2012). As illustrated in Fig. 1, the scope of this article is confined to the structural damage of built infrastructure due to the combined effect of climate- and air pollutants-derived *chemical pathways* and *changing environmental conditions*. Deterioration and blackening of the *historical buildings* are very briefly described for the completeness of the article, given that a wealth of published literature currently exists (see Section 3.1.5). In what follows, the term ‘built infrastructure’ refers to buildings (including heritage) and transport infrastructure (roads, railway tracks, bridges, tunnels, airports, sea ports, earthworks) whereas ‘climate-derived parameters’ refer to temperature, relative humidity, winds, and precipitation.

Impacts of *chemical pathways* are seen in terms of deterioration and blackening of *building materials*. Such impacts are chronic and generally take place over a long-period of time (see Section 3). Quantification of material loss is generally carried out through the dose-response functions (DRFs) which relate climate parameters with the concentrations of air pollutants (Kucera and Fitz, 1995). The major pollutants used as a variable in DRFs are sulphur dioxide (SO₂), ozone (O₃), nitrogen dioxide (NO₂) and particulate matter (PM) (see Table 1). Past studies have made material loss estimations for varying changes in climate parameters and ambient concentrations of air pollutants

1 76 (Section 3). For example, Screpanti and De Macro (2009) carried out corrosion
2 77 assessments of cultural heritage buildings in Italy. For limestone and copper, they found
3 78 the corrosion rates well above the tolerable levels and suggested a need to reduce
4 79 ambient O₃ concentrations in that region. Tidblad (2012) and Ozga et al. (2011) studied
5 80 air pollution induced atmospheric corrosion of metals in Europe and surface damage to
6 81 modern concrete buildings, respectively. Likewise, other studies have raised concerns
7 82 on response of old and cultural heritage buildings due to rapidly changing climate
8 83 parameters and pollutant concentrations (Brimblecombe and Grossi, 2007; Corvo et al.,
9 84 2010; Haines et al., 2006; Sabboni et al., 2006; Varotsos et al., 2009).

15 85 *Changing environmental conditions* are another route of damaging the structures and
16 86 their *materials*. Such impacts are short-lived, acute and intensive when derived by
17 87 extreme weather conditions (e.g. more frequent heat waves and extreme rainfall) and
18 88 long-lived and slow when derived by changing climatic conditions (e.g. increase in the
19 89 average annual temperature, overall drier summers and wetter winters); such climatic
20 90 changes have been confirmed by climate models (Hulme et al., 2002; IPCC, 2007; Karl
21 91 et al., 2009). Both the changes in long-term average climatic conditions and short-term
22 92 extreme events carry a great potential to affect the sustainability of built infrastructure
23 93 (see Section 4).

30 94 Both the chemical pathways and changing environmental conditions are equally
31 95 important for the safety and economy of the *transport infrastructure*. The vast value of
32 96 transport infrastructure assets shows the risk for large economic losses due to the effects
33 97 of both climate change and extreme weather conditions. For example, the highway and
34 98 railway networks in the UK alone have asset values in excess of £87 and £35 billion,
35 99 respectively (Highways Agency, 2009; Network Rail, 2009). Deterioration caused by
36 100 chemical pathways is likely to be realised over a long-term by increase in the
37 101 deterioration rate of construction materials such as steel, cast and wrought iron,
38 102 concrete, masonry and timber. Conversely, extreme environmental conditions can have
39 103 a significant impact in the short term by disrupting road and rail networks (Booij, 2005;
40 104 Nicholls, 2004; UNEP, 2007) that can lead to noticeable economic losses (Larsen et al.,
41 105 2008). In many cases, elements of the transportation infrastructure (e.g. bridge
42 106 structures) also form part of electricity, telephone, water, gas networks. Therefore, the
43 107 economic cost of transport asset and network failures may extend far wider than the
44 108 boundaries of transportation systems to other forms of critical infrastructure (ICE,
45 109 2009). For instance, the bridge failures that occurred in Cumbria (UK) in 2009 due to
46 110 extreme flooding demonstrated the interdependent nature of critical infrastructure where
47 111 these failures resulted not only in loss of transport connectivity and colossal economic
48 112 losses but also to failure of telecommunications, gas and electricity supplies (Stimpson,
49 113 2009).

1 114 Recent reviews have covered numerous topics such as aviation and ground transport
2 115 impacts on climate change (Fuglestvedt et al., 2010; Lee et al., 2010) and recent studies
3 116 have also assessed the impact of climate change on air quality (Athanasidou et al.,
4 117 2010), public health (Vardoulakis and Heaviside, 2012), and buildings (Karaca, 2012;
5 118 McCabe et al., 2011; Ozga et al., 2011). This is the first dedicated review in our
6 119 knowledge which critically presents the impact of air pollution derived chemical
7 120 pathways and changing environmental conditions on the built infrastructure. There are
8 121 five sections in this article. The first section starts with providing the background
9 122 information on the characteristics of corrosive and GHGs to set the context of the study.
10 123 This is then followed by a detailed discussion on a relatively less discussed topic:
11 124 deterioration of building materials by the combined effects of climate parameters and air
12 125 pollutants. A case study is then presented to demonstrate the inter-variability in results
13 126 produced by various DRFs and their usefulness in estimating damage to various
14 127 building materials. The further section discusses the structural and economic impacts
15 128 caused by the changing environmental conditions. Another case study is included within
16 129 this section to demonstrate the effect of changing environmental conditions on the
17 130 material loss of carbon steel (a widely used material in transport infrastructure and
18 131 structural applications). The last section concludes the topic areas covered and
19 132 highlights the research gaps and future challenges.

133 **2. Sources and emission trends of corrosive and GHGs**

134 This section briefly summarises the sources, emission trends, global warming and
135 corrosive potential of key GHGs (carbon dioxide, CO₂; methane, CH₄;
136 chlorofluorocarbons, CFCs, and sulphurhexafluoride, SF₆, nitrous oxide, N₂O) and
137 corrosive gases (SO₂, O₃ and NO₂) for setting up the scene for further discussion and
138 completeness of the article. The GHGs other than the CO₂ have relatively less
139 atmospheric concentrations but they carry a much higher global warming potential (see
140 Table 2). All the GHGs included in Table 2 may not directly deteriorate the materials of
141 buildings and transport structures but these contribute indirectly to influence global
142 radiation balance (Jacob and Winner, 2009; Ramanathan and Feng, 2009) and hence the
143 climate parameters and concentrations of corrosive gases via chemical transformation
144 (see Table 1 and Section 3). The mean global concentrations of various GHGs and
145 changes in their concentrations over the past decade are presented in Table 2.

146 **2.1. CO₂**

147 As seen in Fig. 2, CO₂ is the most important GHGs with highest growth rate,
148 largest atmospheric concentrations and substantially longer atmospheric life time. This
149 is mainly produced through consumption of fossil fuels in industries and the transport
150 sector, energy consumption in households, deforestation and natural degradation of
151 biomass sources due to oxidation of carbon compounds in marshes and forests (Climate
152 Change, 2007). Annual emissions of CO₂ have grown by ~80% from 21 to 38

153 gigatonnes between 1970 and 2004, representing ~77% of total anthropogenic GHG
154 emissions in 2004 (Climate Change, 2007). Currently, the transportation sector is
155 responsible for a large share (20–25%) of worldwide man-made CO₂ emissions
156 (Fuglestvedt et al., 2008) that could increase to 30–50% by 2050 (Kumar et al., 2010;
157 Nakicenovic et al., 2000). Besides playing a key role in changing climate parameters
158 (Jacob and Winner, 2009; Ramanathan and Feng, 2009), carbonation caused by
159 atmospheric CO₂ to concrete structures is one of the major physicochemical processes
160 (Section 3.1.2) which can compromise the service life of reinforced concrete structures
161 such as buildings and bridges (Peter et al., 2008; Tonoli et al., 2010).

162 **2.2. CH₄**

163 CH₄ has the second greatest effect on global climate after CO₂ (Table 2). Main
164 sources of CH₄ includes anaerobic degradation of organic matter in rice fields, natural
165 wet lands, landfills, digestive tract of cattle, production and use of oil and natural gas
166 and incomplete burning of organic material. The global atmospheric concentration of
167 CH₄ has increased over 2-fold from a pre-industrial value of about 715ppb to 1732ppb
168 in the early 1990s, and to 1774ppb in 2005, mainly due to increased agricultural
169 activities and fossil fuel use (Climate Change, 2007). Like CO₂, CH₄ does not corrode
170 the material directly, but influences the chemistry of O₃ (Section 2.8) which is one of
171 the most corrosive gases and the climate parameters through radiative forcing.

172 **2.3. N₂O**

173 N₂O is the fourth largest single contributor to positive radiative forcing, and
174 serves as the long lived atmospheric tracer of the human perturbations of the global
175 nitrogen cycle (IPCC, 2007). This is produced from the burning of biomass and nitrogen
176 rich fuels (especially coal), break down of nitrogen fertilisers in soil, livestock wastes
177 and nitrate contaminated ground water. The global atmospheric N₂O concentration
178 increased from a pre-industrial value of about 270ppb to 319ppb in 2005, and a further
179 increase by 35–50% up to 2030 is expected due to increased nitrogen fertilizer use and
180 increased animal manure production (Climate Change, 2007). The N₂O does not
181 contribute to material corrosion directly, but its positive contribution to radiative forcing
182 influences the environmental conditions.

183 **2.4. CFCs**

184 CFCs contribute to global warming by absorbing long-wave radiation (Fisher et
185 al., 1990). These come into the environment through leaking air conditioners,
186 refrigerators, evaporation of industrial solvents, production of plastic foams, aerosols
187 and propellants. Their global mean atmospheric concentrations have increased from a
188 near-zero pre-industrial background to about 251 ppt (CFC-11), 538 ppt (CFC-12) and
189 78 ppt (CFC-113) in 2005 (IPCC, 2007). When ultraviolet radiation strikes the CFCs
190 molecules, carbon-chlorine bond is broken and chlorine atoms are produced. A single

1 191 chlorine atom in the stratosphere can destroy 100,000 molecules of O₃ over its natural
2 192 lifetime. Thus the CFCs play a crucial role for changing the radiation balance and hence
3 193 the climate parameters.

6 194 **2.5. SF₆**

8 195 SF₆ is an extremely stable atmospheric trace gas that is produced entirely through
9 196 anthropogenic emissions. About 80% of the SF₆ produced worldwide is used in and
10 197 released from electrical equipment, predominantly from gas insulated switchgear (Maiss
11 198 et al., 1996). SF₆ emissions are also one of the six GHGs targeted for reduction under
12 199 the Kyoto Protocol. Studies show an increase in mixing ratios from near zero in the
13 200 1970s to a global mean value of 4.1 ppt in 1998 (IPCC, 2007) and 6.3 ppt by the end of
14 201 2007 (Levin et al., 2009), showing about 54% increase in 2007 over the 1998 levels. In
15 202 our context, the role of SF₆ is in the form of influencing the climate parameters through
16 203 disturbing the radiation balance.

21 204 **2.6. SO₂**

23 205 SO₂ is predominantly produced through sulphur-led anthropogenic activities
24 206 occurring in industries (e.g. power generation using coal in refineries, cement, brick and
25 207 ceramic), through waste incineration, road transport emissions and mineral processing
26 208 like the ferrous metal production (Ramanathan and Feng, 2009). SO₂ shows cooling
27 209 effects on climate in low or moderate emission conditions (Ward, 2009), but it is crucial
28 210 for deteriorating the materials used in buildings and transport infrastructure (see Section
29 211 3). SO₂ in the atmosphere is oxidised to particle phase and gas species after reactions
30 212 with other atmospheric species, which then further oxidised to its acid forms (e.g.
31 213 sulphurous acid, H₂SO₃, sulphuric acid, H₂SO₄) on the surface of materials. These acids
32 214 react with the building materials to damage them. Corrosion impacts of SO₂ are greatly
33 215 influenced by the temperature and relative humidity. For example, Noah's Ark (2006)
34 216 reported that corrosion of materials is expected to increase in the Northern part of the
35 217 Europe and to decrease in the southern part even if atmospheric levels of SO₂ remain
36 218 constant. Emissions of acidic air pollutants such as the SO₂ are likely to decrease in
37 219 future in many parts of the world, most certainly in European countries, as a
38 220 consequence of stricter regulations (Brimblecombe and Grossi, 2007). This is likely to
39 221 lead to an overall improvement in deterioration rates of various materials attributed to
40 222 acid deposition. More precisely, current observations reveal a continuous decrease by
41 223 70–90% in SO₂ emissions since the 1970s in Europe, following a series of control
42 224 measures such as the decreasing sulphur contents in diesel fuel and implementation of
43 225 air pollution control measures in power plants (Fowler et al., 2007). A recent study by
44 226 Kuribayashi et al. (2012) summarised the long-term trends of SO₂ emissions and
45 227 deposition in East Asia, indicating contrasting trends to those reported for European
46 228 environments. For example, China accounts for 64–71% of the total Asian SO₂
47 229 emissions; a rapid increase in its emissions were noted between 1980s to the mid-1990s,

1 230 9–17% decrease between 1996 and 2000, and again a sudden increase of about 61%
2 231 between 2000 and 2006, followed by a 9.2% decrease during 2006–2010 (Lu et al.,
3 232 2011). As opposed to China, India’s national emissions of SO₂ have increased by 70%
4 233 in 2010 (8.8 Tg) from the 1996 levels (5.2 Tg) and the current emission trend is upward
5 234 (Lu et al., 2011).

9 235 **2.7. NO_x**

10 236 NO_x is generally taken as the sum of NO and NO₂. This is a corrosive gas which
11 237 is oxidised to its acid form (e.g. nitrous acid, HNO₂, nitric acid, HNO₃) on the surfaces
12 238 of materials. These acids are responsible to much of the damage to building materials
13 239 (Section 3.1.1). This also play an important role in the formation of tropospheric O₃
14 240 (Lewne et al., 2004; Nagpure et al., 2011). A dominant fraction of NO_x is produced
15 241 from the combustion activities at high temperatures, such as in engines of motor
16 242 vehicles. The situation of NO_x emissions in Europe is different to those in, for example,
17 243 Asia or Latin America where the emissions are appearing to be increasing due to less-
18 244 stringent policy regulations (Vestreng et al., 2009). Recent studies have found a
19 245 significant decrease in NO_x concentrations in compliance with the EU standards and
20 246 directives at roadside sites in Europe (Mavroidis and Chaloulakou, 2011). This decrease
21 247 is, though, not as fast as foreseen by the directives regulating vehicle emissions. Despite
22 248 the overall NO_x reductions in Europe, a corresponding decrease in the ambient NO₂
23 249 concentrations has not yielded and its exceedances over the limit values are being
24 250 reported in many urban locations across Europe (Mavroidis and Chaloulakou, 2011).
25 251 This is primarily due to significant increase in the NO₂/NO_x ratio, highly non-linear
26 252 dependency of secondary NO₂ (contributing ~70% of total) on NO_x, and the increasing
27 253 NO₂ ratio in late diesel engine technology vehicles (Vestreng et al., 2009). However,
28 254 ever growing stricter emission standards in Europe and both the in-cylinder and after-
29 255 treatment control measures are likely to bring reduction in NO_x and primary NO₂
30 256 emissions in future.

41 42 257 **2.8. O₃**

43 258 O₃ is a secondary pollutant and its concentrations depend on the atmospheric
44 259 chemistry and emissions of its anthropogenic precursors (e.g. NO_x, CO, CH₄ and
45 260 volatile organic compounds, VOCs) that originate from fossil fuel combustion and
46 261 natural sources such as vegetation. The O₃ is generated through the photochemical
47 262 oxidation of its precursors by the hydroxyl radical (OH) in the presence of reactive NO_x
48 263 (Jacob and Winner, 2009). OH radicals are formed through atmospheric oxidation of
49 264 water vapour and cycles in the atmosphere with other hydrogen oxide radicals. The
50 265 sources of O₃ precursors are generally located in the boundary layer where the lifetime
51 266 of O₃ is of the order of days (Jonson et al., 2005; Nagpure et al., 2011). Major sinks of
52 267 ground-level O₃ are dry deposition on vegetation and photolysis which occurs in the
53 268 presence of water vapour (Jacob and Winner, 2009). O₃ is a powerful oxidiser which

1 269 exerts a direct corrosive effect on various materials that are even measurable at as low
2 270 as 20 ppb atmospheric concentrations (CC Report, 2003). Despite reductions in the
3 271 emissions of O₃ precursors, predictions associated with global climate change have
4 272 shown that ground-level O₃ concentrations are expected to increase in future years
5 273 (Zeng et al., 2010). For example, background concentrations of ground-level O₃ in
6 274 northern mid-latitudes have almost doubled to 35–40 ppb since the 19th century;
7 275 anthropogenic emissions contribute half to two third of these emissions (Garthwaite et
8 276 al., 2009). At the same time, recent studies also conclude that magnitude and origin of
9 277 ground-level O₃ trends in Europe are not completely understood, and that the local
10 278 trends can be dominated by the changes in emissions of O₃ precursors (Jonson et al.,
11 279 2005). O₃ concentrations in the atmospheric boundary layer result in continuous
12 280 oxidation of polymers and building materials used in modern construction, and may also
13 281 enhance the production of acids after reacting with SO₂ and NO_x (Brimblecombe and
14 282 Grossi, 2007). As discussed in Section 3, numerous studies have reported impact of
15 283 ground-level O₃ on various building materials and its increasing importance in future
16 284 years (CC Report, 2003; Noah's Ark, 2006; Prather et al., 2003; Screpanti and De
17 285 Marco, 2009).

27 286 **3. Chemical pathways: material deterioration of built infrastructure**

28 287 This section firstly describes the chemical sensitivity of materials used in the built
29 288 infrastructure. This is then followed by a comprehensive review of available DRFs
30 289 which are presented in Table 3. Further section assesses the practical applications and
31 290 limitations of the DRFs with the help of a case study.

32 291 **3.1 Chemical sensitivity of various building materials**

33 292 Atmospheric corrosion of materials is a cumulative and irreversible process that
34 293 may occur even in the absence of air pollutants (Helcher et al., 1991). Structures are
35 294 predominantly made of steel, concrete, brick masonry, stone and wood materials and
36 295 have particular sensitivity towards various chemical processes as described below.

37 296 **3.1.1 Oxidation sensitive materials**

38 297 *Steel* and *stone* materials are largely influenced by oxidation processes led by
39 298 SO₂, NO_x and O₃ (see Table 3). The SO₂ can deposit on the surface of materials and
40 299 oxidise into SO₄ (Corvo et al., 2010). After reacting with the water droplets present in
41 300 the atmosphere, this then forms H₂SO₄ which corrodes these materials by coming in
42 301 contact through rain (Livingstone, 1992). NO_x has similar characteristics to form HNO₃
43 302 at the surface of the material and within the atmosphere to affect these materials. O₃ is a
44 303 powerful oxidiser and its effect on steel and stone is measurable even at very low (20
45 304 ppb) atmospheric concentrations (CC Report, 2003). A study by Lee et al. (1996)
46 305 assessed the potential damage by O₃ to the materials in the UK as £170–£345 million
47 306 yr⁻¹. This includes damage to surface coatings and elastomers, and the cost of anti-

307 ozonant protection applied to rubber goods. The effects of O₃ on the costs of repainting
308 were estimated in the range of £0–£60 and £0–£182 million yr⁻¹ for a change in O₃ from
309 15 to 20 ppb and 15 to 30 ppb, respectively.

3.1.2 Carbonation sensitive materials

310 *Plain or reinforced concrete* is semi impermeable, one of the oldest, most durable,
311 widely used composite, but a carbonation sensitive building material. Past studies have
312 examined qualitative impacts of air pollution and changing environmental conditions on
313 reinforced concrete deterioration but methods for quantitative estimations are still an
314 open area for research (Ozga et al., 2011). At present, there are no DRFs available for
315 concrete, clearly because it is not a single material but a composite. Reinforced concrete
316 is affected by chloride attack, CO₂-induced carbonation and freeze-thaw (thermal
317 shocks) processes, mainly caused by climate parameters (Jacobsen et al., 1995; Shang et
318 al., 2009). While high presence of waterborne salts, particularly chloride ions, initiates
319 the chloride attack to reinforcement, carbonation is characterised by the introduction of
320 factors that destroy the protective passive layer of reinforcement. This passivating
321 environment for the reinforcement is destroyed by multi-pollutants, derived mainly by
322 oil and coal combustion processes, such as NO_x/HNO₃. CO₂/H₂CO₃ (carbonation), or
323 SO₂/H₂SO₄ (sulphation) (Ozga et al., 2011). The latter influences the alkalinity of
324 concrete and converts calcium aluminate into ettringite and gypsum (CaSO₄.2H₂O), and
325 the gypsum thus formed occupies a larger volume than the original concrete and lead to
326 surface deterioration. This is an important damage process to affect concrete structures
327 in SO₂ rich urban environments (Marinoni et al., 2003). This is particularly the case
328 with cities in developing countries (e.g. India; see Section 2.6) where SO₂ levels are yet
329 increasing compared with decreasing SO₂ emission trends in European and North
330 American cities, mainly due to declining sulphur contents in diesel fuel (Lu et al., 2011;
331 Kuribayashi et al., 2012; Stern, 2005; Kumar et al., 2011). Likewise, atmospheric CO₂
332 diffuses through the unsaturated concrete pores, reacts with carbon solutes and then
333 forms a mildly acidic solution (Peter et al., 2008). This results in a drop in pH value of
334 inside concrete environment from about 13. When this goes below about 10.5,
335 carbonation starts to erode the concrete material and exposes the reinforcing steel within
336 it. Similar principle applies to plain concrete structures. External facades of these
337 structures are particularly vulnerable to such phenomena because of increased
338 susceptibility of the condensation of water. However, plain concrete material is more
339 resistant, due to the absence of reinforcement, which is more sensitive to corrosion;
340 carbonation for instance is not such a serious problem as in the case of reinforced
341 concrete.

343 Peng and Stewart (2008) analysed the impacts of increasing concentrations of CO₂ on
344 carbonation of reinforced concrete. They concluded that the effect of carbonation is
345 much higher on high water-cement (w/c) ratio concrete as opposed to low w/c ratio

1 346 concrete, with distance between carbonation front and reinforced steel bar being a key.
2 347 They also developed a method for time dependent reliability analysis to calculate
3 348 probabilities of corrosion initiation, mean proportions of corrosion damage and
4 349 probabilities of structural collapse when the CO₂ concentration increases with time over
5 350 the next 100 years. They found that probability of corrosion initiation to reinforced
6 351 concrete structures can be up to 720% higher than a scenario based on maximum
7 352 mitigation of CO₂ emissions. The worst emissions scenario increased the likelihood and
8 353 extent of corrosion damage by 540% when compared to the structural reliability for the
9 354 best mitigation scenario (Peng and Stewart, 2008). Other studies have also proposed
10 355 diffusion models for carbonation (CEB, 1997; Kersner et al., 1996) and found their
11 356 modelled results most sensitive to increase in atmospheric CO₂ (Stewart et al., 2011).
12 357 Detailed investigation of these models is out of the scope of our review, but further
13 358 information on this topic can be found in Papadakis et al. (1992), Yoon et al. (2007),
14 359 Peter et al. (2008), Bastidas–Arteaga et al. (2010), Stewart et al. (2011), and references
15 360 therein.

23 361 **3.1.3 Soiling / weathering sensitive materials**

24 362 Similar to stone, *brick material* is prone to weathering or soiling process. The
25 363 process is caused by changing environmental conditions such as the variations in
26 364 temperature and relative humidity, wind speed, freezing and thawing, extraction by rain
27 365 and snow melt water, acid rain and microbial activities (Hirsch et al., 1995). Acid rain is
28 366 one of the most prominent chemical deterioration processes for brick masonry structures
29 367 because of: (i) the susceptibility of bricks to acid rain through the selective dissolution
30 368 of their glassy phase, (ii) the reactions within the calcareous components of mortar
31 369 affecting its strength, and (iii) migration of soluble salts resulting from the above
32 370 reactions (within the solution with rain water or condensed moisture) through the
33 371 porous matrix of the masonry, and finally (iv) evaporation of water leaving salt
34 372 deposited on the brick surfaces and such repeated dissolution and recrystallisation
35 373 leading to the mechanical disruption of the masonry structure (Charola and Lazzarini,
36 374 1986). Other important mechanisms are frost and freeze–thaw actions, but these are
37 375 mainly derived by extreme or changed environmental conditions (Kvande and Lisø,
38 376 2009; Nijland et al., 2009). Rather limited information is available on brick material and
39 377 future investigations are needed to analyse the detailed impacts of changing climate and
40 378 air pollutants on such structures.

49 379 **3.1.4 Moisture sensitive materials**

50 380 Little is known about the deterioration of *wood or timber* through airborne
51 381 chemical species. There are however evidences relating their deterioration with climate
52 382 parameters, mainly atmospheric moisture. The principal mechanism for damaging such
53 383 open–air structures is attack by wood pests, especially fungi (Herlyn and Mehlhorn,
54 384 1999; Nijland et al., 2009). Chances of fungal attacks are even more in high

1 385 precipitation areas. This process is mainly driven by moisture penetration to woods and
2 386 ambient temperature, and is generally estimated through ‘Scheffer Climate Index
3 387 (SCI)’. The higher the SCI the greater is the decay hazard i.e. >65 (severe), 35–65
4 388 (moderate) (Scheffer, 1971). Fungal growth starts occurring when moisture penetration
5 389 goes past the critical value which is ~20% as estimated by Noah’s Ark (2006). Wood–
6 390 inhabiting fungi may grow in a temperature range of about +3 °C to about +40 °C with
7 391 an optimum temperature of about +25 °C (Hof, 1981). The Noah's Ark (2006) report
8 392 concluded that wood deterioration due to fungal growth can increase up to 100% in high
9 393 precipitation areas (e.g. Scandinavian countries and Northern Russia). A recent study by
10 394 Brischke and Rapp (2008) established DRFs for Scots pine sapwood and Douglas fir
11 395 heartwood for up to 7 years exposure in 27 different European test sites. They found
12 396 that a traditional ‘SCI’ is not an appropriate tool for estimating site specific decay
13 397 potential of woods. Rather, they established DRFs, predominantly based on moisture
14 398 content and wood temperature, for predicting service life of woods (see Table 3).
15 399 However, these results were for a limited set of experimental conditions and detailed
16 400 studies establishing relationship between moisture content and the amount and duration
17 401 of rainfall under different exposure situation are still required (Brischke and Rapp,
18 402 2008). Furthermore, there are currently a negligible number of studies available relating
19 403 wood deterioration with both climate parameters and air pollutants. Such studies are
20 404 needed to better understand this problem.

31 405 Another type of material, which is sensitive to moisture–induced deterioration, is Fibre
32 406 Reinforced Polymer (FRP) composites, used widely for strengthening as well as for the
33 407 construction of light-weight structures. Absorption of moisture leads to hydrolysis,
34 408 plasticization and saponification on the resin which may cause irreversible changes in
35 409 the polymer structure and loss of integrity of the fibre–matrix structure. The polymer
36 410 matrix present in composite materials is also prone to degradation initiated by
37 411 ultraviolet (UV) radiation, temperature and high pH environments (Chin et al., 1997).
38 412 Exposure to sub–zero temperatures can result in matrix hardening, micro–cracking and
39 413 degradation and freeze–thaw cycles can lead to accelerated degradation of the material.
40 414 Exposure to UV radiation leads to surface deterioration which adversely affects
41 415 mechanical properties and can increase moisture ingress in the deteriorated regions.
42 416 Although the effect of environmental conditions on the durability of composite
43 417 materials are qualitatively known, as discussed above, actual data on the durability of
44 418 composite materials is sparse due to the lack of long-term experiments to different types
45 419 of atmospheric and environmental conditions (Karbhari et al., 2003).

420 **3.1.5 Deterioration and blackening of buildings**

421 Of the notable research efforts carried out in recent years on this topic has been
422 the work by Brimblecombe, Grossi, McCabe and their co–workers, besides the
423 researchers from the ICP projects (Tidblad et al., 2001), Noah’s Ark (2006) and

MULTI-ASSESS (Kucera et al., 2007). The impact of changing environment on buildings can occur in the form of deterioration and blackening of building stones, driven by processes such as freeze–thaw cycles, wind driven rain, humidity cycles and salts, gas and particle concentrations, pH of precipitation, and the water table level (Brimblecombe and Grossi, 2007; Karaca, 2012; Ozga et al., 2011). Blackening of buildings materials due to accumulation of PM containing dark elemental carbon is a common problem (Haynie, 1986). Grossi et al. (2008) concluded that recession rates of architectural limestone buildings in European cities will remain largely unchanged over the coming century due to the continuously declining level of air pollutants despite the changes in climate parameters. However, a dramatic change can occur in blackening pattern of buildings due to new climate regimes and a somewhat different trend may emerge. For instance, reduced emissions of PM and increase in seasonal rainfall may help in self–cleaning of buildings, but probably at the expense of encouraging micro–organisms growth that may result in *yellowing* of buildings (Noah's Ark, 2006). Recent studies have also indicated the biological *greening* of natural stone buildings. The predominant reasons quoted for such effects is algal growth in response to wetter exposure conditions, possibly in combination with reduced atmospheric SO₂ and an increase in atmospheric nitrogen (NO_x) from vehicular pollution at some locations (McCabe et al., 2011). A wealth of literature is available on the impacts of climate change on natural building stones and therefore is covered here briefly for the completeness of the article.

3.2 DRFs

Atmospheric pollutants influence the global climate by regulating the radiation budget (Isaksen et al., 2009) and act together with a range of climate parameters to damage the building materials (Texte 24/99, 1999). The DRFs serve as a tool to quantitatively assess material damage by combining both the climate parameters and pollutant concentrations using the general expression as shown in Eq. (1) (Kucera and Fitz, 1995; Tidblad, 2007; Tidblad et al., 2001):

$$ML = f_{\text{dry}}(T, Rh, [\text{GHG and corrosive gases}]) \times t^m + f_{\text{wet}}(Rn, [\text{ions}]) \times t^n \quad (1)$$

where ML, T, Rh and t refer to material loss ($\mu\text{m yr}^{-1}$ or g m^{-2}), temperature ($^{\circ}\text{C}$), relative humidity (%) and exposure duration (year), respectively. Rn is precipitation (m yr^{-1}); ions are concentrations of $[\text{H}^+]$ that are derived from the pH of rain; m and n are empirical constants. The first part of Eq. (1) indicates material loss due to dry corrosion while the second part refers to wet corrosion. The above expression is modified by numerous authors for establishing a number of empirical relationships presented in Table 3.

As seen from the DRFs in Table 3, SO₂ and O₃ are important corrosive gases. SO₂ shows a non–linear relationship (exponent generally less than unity) with corrosion and

its corrosive effect is maximum at a temperature of about 9–11°C (Kucera et al., 2007). Although, SO₂ is no longer a dominant pollutant in most parts of the developed world, concentrations of ground-level O₃ still remain a major concern (see Sections 2.6 and 2.8). Relative humidity, H⁺ ions in rain and temperature are important for corroding most metals while Cl⁻ deposition plays an important role in corroding carbon steel, aluminium and cast bronze (Table 3). It should be noted that concrete is not included in the listed materials due to the lack of available DRFs.

New versions of international standards have recently been published (BS EN ISO 9223, 2012; BS EN ISO 9224, 2012) for classifying, determining and estimating the corrosivity of metals and alloys. These standards are widely adopted throughout the world for the purposes of corrosion assessment of materials. BS EN ISO 9223 (2012) provides DRFs for calculating the first-year corrosion rate of carbon steel, zinc, copper and aluminium materials; their summary is presented in Table 4. The ISO-proposed DRFs are based on the previous research and hence show a similar form to those listed in Table 3 (i.e. material loss is a function of atmospheric pollutant concentrations and climate parameters).

For long-term estimation of corrosion loss, the effect of changes in atmospheric pollutant concentrations as well as in environmental conditions (climate parameters) is needed to be taken into account. BS EN ISO 9224 (2012) provides such relationships for estimating the total corrosion loss, D (µm), of various metals:

$$D = r_{corr} t^b \quad \text{for } t < 20 \text{ years} \quad (2)$$

$$D = r_{corr} [20^b + b(20^{b-1})(t - 20)] \quad \text{for } t > 20 \text{ years} \quad (3)$$

where r_{corr} is the first-year corrosion rate in µm yr⁻¹ (see Table 4 for detailed expressions), t is the exposure time in years, and b is a metal-environment-specific time exponent (i.e. $b=0.523$ for carbon steel; $b=0.813$ for zinc, $b=0.667$ for copper, $b=0.728$ for aluminium). By looking at Eq. (2) and (3), it can be seen that changes in atmospheric pollutant concentrations and climate parameters can only be captured through the first-year corrosion rate, r_{corr} . Therefore, it can be argued that the BS EN ISO 9224 (2012) functions are not well-suited for taking into account gradual changes of the above parameters over time.

3.2.1 Case study for analysing the inter-comparability of DRFs

For analysing the inter-comparability of various DRFs, a case study is conducted on 4 different materials (limestone, carbon steel, zinc and copper) and the results are summarised in Fig. 3. The DRFs employed to compute the losses for different materials are generally those currently available in the published literature (see Table 3). All the materials selected for this case study have their own significance. For

1 498 instance, limestone is often used in heritage buildings whereas carbon steel is widely
2 499 used in built infrastructure for the construction of residential, commercial and industrial
3 500 buildings as well as bridge structures. Zinc and copper are widely used as roofing and
4 501 cladding materials in buildings. These materials are assumed to be exposed in
5 502 unsheltered environmental conditions in London. For the purposes of the inter-
6 503 comparability analyses, we have adopted the air pollution and climate data from
7 504 Brimblecombe and Grossi (2008) for the years 1990 and 2010. PM₁₀ data is taken from
8 505 Fuller and Green (2006) and IAEI (2006). Dry deposition velocities for SO₂ and HNO₃
9 506 are assumed to be 0.38 and 0.32 cm s⁻¹, respectively (Sabboni et al., 2006). The
10 507 subsequent paragraphs discuss the results obtained from this case study and the further
11 508 Section 3.2.2 provides summary and critical discussion of the results.

12 509 The common observation from Fig. 3 is that material losses averaged over the values
13 510 computed by various DRFs for an individual material have decreased in 2010 from the
14 511 1990 levels, mainly due to reduced concentrations of SO₂, O₃ and PM₁₀. These were
15 512 found to be fallen by 3.2, 9.8, 4.3 and 19.2% in 2010 from the 1990 levels for carbon
16 513 steel, limestone, zinc and copper, respectively. However, the main objective of these
17 514 computations is not to demonstrate the long-term temporal trend of material loss, as
18 515 recently presented by studies elsewhere (Brimblecombe and Grossi, 2008; Graedel and
19 516 Leygraf, 2001; Grossi et al., 2008; Tidblad, 2012), but to analyse the comparability of
20 517 results obtained from various DRFs for identical input parameters and compare the
21 518 estimated values with those published in literature.

22 519 Inter-comparison of modelled results using different DRFs show reasonably close
23 520 values to each other for all materials. The coefficient of variances (CoV) were computed
24 521 in each case to assess the inter-comparability of different DRF results. For instance, 4
25 522 different DRF models were applied to measure the surface recession of *limestone* (Fig.
26 523 3a). These produced the highest CoVs among all materials considered as 0.25 and 0.30
27 524 for 1990 and 2010, respectively. This variability is mainly dominated by the Lipfert
28 525 (1998) function which produces the largest surface recession compared with other 3
29 526 functions used, because of its relatively higher weightage for precipitation (i.e. karst
30 527 effect). As seen in Table 3, the structure of the Lipfert function recognises the three
31 528 mechanisms for material loss: (i) karst effect (CO₂ weathering), (ii) acid rain effect, and
32 529 (ii) dry deposition (Brimblecombe and Grossi, 2008). The current estimates are for
33 530 “clean precipitation” and ignore the maritime influence on the karst effect, given that
34 531 deposition of sea salt aerosol has maximum effect within the first 100 m (Bonazza et al.,
35 532 2009). The recession estimates are found to be dominated by the karst effect set as 18.8
36 533 µm per m precipitation per year, which contributed to over 90% of total estimated
37 534 surface recession (Fig. 3a). However, the work by Bonazza et al. (2009) indicates that
38 535 the choice of the above value is acceptable because (i) natural recession rate becomes
39 536 large only when the porosity of material exceeds ~25%, and (ii) most of building stones

1 537 are of medium porosity, thus falling close to the value adopted by Lipfert. Otherwise,
2 538 the remaining three models for limestone produce results nearly close to each other (i.e.
3 539 8.37 ± 1.06 and $7.41 \pm 1.16 \mu\text{m yr}^{-1}$ for 1990 and 2010, respectively). The main difference
4 540 between the Lipfert and other functions are that Tidblad et al. (2001) takes temperature
5 541 into account, Kucera et al. (2007) omits temperature and introduce relative humidity and
6 542 PM_{10} , and Screpanti and Macro (2009) include temperature, relative humidity and PM_{10} .
7 543 All these three input variables are not part of the Lipfert function which provides
8 544 considerable weight to karst effect and underestimates surface recession caused by
9 545 atmospheric pollutants (Bonazza et al., 2009; Delalieux et al., 2002). The trend of
10 546 getting the largest recession rates by the Lipfert (1998) function is consistent with the
11 547 results obtained by Grossi et al. (2008) for Oviedo (Spain), Paris (France) and Prague
12 548 (Czech Republic) where they applied Lipfert (1998), ICP (Tidblad et al., 2001) and
13 549 MULTI-ASSESS (Kucera et al., 2007) DRFs. Our annual recession rates compare well
14 550 to those estimated by Kucera et al. (2007) for London for the years 1997–2001 ($\sim 8 \mu\text{m}$
15 551 yr^{-1}) using their function for multi-pollutant situation and are on the upper end of the
16 552 tolerable levels ($8 \mu\text{m yr}^{-1}$) for limestone (Kucera et al., 2005).

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18 554 The DRFs developed by the Noah's Ark (2006) and MULTI-ASSESS (Kucera et al.,
19 555 2007) projects were used for estimating surface recession of *carbon steel*. These
20 556 produced remarkably less variability in results for both the years (i.e. CoV = 0.03 for
21 557 1990 and 0.04 for 2010; see Fig. 3b). Given that salt aerosol deposition generally does
22 558 not exceed 5 km from the shore with the maximum deposition within the first 100 m
23 559 (Bonazza et al., 2009) and that of the unavailability of sea salt data, we ignored Cl^-
24 560 deposition terms from both functions. Results indicate SO_2 as the most sensitive term,
25 561 followed by PM_{10} , in these functions dictating the total recession values. Our estimates
26 562 are on the higher end to those reported by the United Nations report (UN, 2008) as ~ 26
27 563 and $28 \mu\text{m yr}^{-1}$ in London during 1990 and 2010, respectively. This is because our SO_2
28 564 concentrations (20 and $17 \mu\text{g m}^{-3}$) are up to 2 times larger than the constant
29 565 concentrations ($10 \mu\text{g m}^{-3}$) assumed by the UN (2008) estimates. The average surface
30 566 recession was found to be about 1.60 and 1.50 times larger in 1990 and 2010,
31 567 respectively, than the tolerable values ($20 \mu\text{m yr}^{-1}$) for carbon steel (Kucera et al.,
32 568 2005). These observations clearly suggest a need to control the emissions of PM_{10} for
33 569 reducing the carbon steel recession to tolerable levels.

34 570 The ICP (Tidblad et al., 2001) and MULTI-ASSESS (Kucera et al., 2007) DRFs are
35 571 used for estimating corrosion losses to *zinc* and *copper*. As expected, these functions
36 572 produced consistently comparable estimates for both the materials as evident from close
37 573 to zero CoV values (Figs. 3c and d). Consistent with previous cases, the material loss
38 574 for both zinc and copper is higher in 1990 compared with 2010, mainly because of
39 575 reduced SO_2 and O_3 concentrations. Our estimates for zinc (9.29 and 8.90 g m^{-2} for
40 576 1990 and 2010, respectively) and copper (9.54 and 8.00 g m^{-2} for 1990 and 2010,

1 577 respectively) were found to be comparable to those reported by Kucera et al. (2007) and
2 578 Kucera et al. (2005) during 1997–2001 (i.e. $\sim 8 \text{ g m}^{-2}$) and 2002 (i.e. up to $\sim 7 \text{ g m}^{-2}$) for
3 579 London, respectively.
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6 580 **3.2.2 Discussion: practical application and limitations of the DRFs**

7 581 DRFs should ideally be suitable for mapping any area with increased risk of
8 582 corrosion and allowing costing of corrosion damage, as applied by recent studies
9 583 (Karaca, 2012; Tidblad, 2012). There are a number of functions available in the
10 584 literature based on linear or non-linear empirical equations (Table 3) but they should be
11 585 generalised cautiously for untested conditions. This is because their outputs may differ
12 586 depending on the assumptions, mainly driven by precipitation, concentrations of
13 587 pollutants and relative humidity (Grossi et al., 2008). This can be argued that which
14 588 function should be selected when more than one DRF is available for a particular
15 589 material? The key in such a case is to carefully assess the climate and pollution
16 590 conditions in which the DRFs are developed. For instance, the DRFs developed before
17 591 or in early 2000 (e.g. through the ICP; Tidblad et al., 2001) could work well for SO_2
18 592 dominating environments such as in Asian cities (see Section 2.6). Recently developed
19 593 functions (e.g. through MULTI-ASSES project; Kucera et al., 2007) would be more
20 594 appropriate for the multi-pollutant environmental conditions, generally dominated by
21 595 the road vehicle emissions. When available DRFs were used on selected materials in
22 596 Section 3.2.1, all the functions provided reasonably close values (Fig. 3). However, one
23 597 limitation of this case study is that only modelled results are compared with each other
24 598 and these are not evaluated against the measured values for real structures. Furthermore,
25 599 most of the available functions are developed by exposing the specimens in
26 600 sheltered/unsheltered conditions, but not the real structure. Previous research has shown
27 601 that the orientation of the exposed specimen can affect the amount of deterioration
28 602 considerably (Knotkova et al., 1982; Coburn et al., 1995). Further concern could include
29 603 the reliability of long-term future projections using the available DRFs under the
30 604 changing environmental conditions for real structures, especially at the locations other
31 605 than where they were originally developed. There appears to be a clear need for
32 606 developing more robust and theoretical functions, and their performance evaluation
33 607 against the measured data, which could be adopted for a variety of environmental and
34 608 pollution conditions (Grossi et al., 2008).
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49 609 **4. Changing environmental conditions: structural and economic** 50 610 **impacts on the integrity of built infrastructure**

51 611 In addition to the chemical pathways described in the previous sections, changing
52 612 and extreme environmental conditions arising from climate change may also have the
53 613 potential to affect the structural integrity of the building and transport assets. Table 5
54 614 summarises such potential impacts due to the following projected global climatic
55 615 conditions: (i) warmer and wetter winters; an average global rise in temperature that can
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1 616 range from 1.4 to 5.8 °C between 1990 and 2100 (Climate Change, 2007; IPCC, 2007),
2 617 (ii) increase in extreme rainfall amounts and more frequent extreme precipitation events
3 618 (IPCC, 2007; TRB, 2008), (iii) hotter and drier summers with more frequent and
4 619 extreme high temperatures (Arkell and Darch, 2006; IPCC, 2007), and (iv) rise in
5 620 global sea level ranging from 18 to 59 cm between 1990 and 2100 with an increased
6 621 risk of tidal surges (relative mean sea level rise) (IPCC, 2007).

10 622 The recent UK Climate Change Risk Assessment report has identified the most
11 623 significant risks due to changing environmental conditions on a wide range of sectors
12 624 ranging from agriculture and forestry, business, health and wellbeing to buildings and
13 625 infrastructure and the natural environment (CCRA, 2012). Risks have been prioritised
14 626 according to their criticality and potential impact. The main structural risks that have
15 627 been identified for buildings are damage to properties due to flooding and coastal
16 628 erosion as well as subsistence (settlements). On the other hand, the main risks for
17 629 transport infrastructure have been identified as flooding of roads and railways, scouring
18 630 of road and rail bridges, and landslips.

24 631 Significant *economic impacts* are estimated due to these climate changes. For example,
25 632 the total value of assets of 136 port cities worldwide, which have a population over 1
26 633 million, exposed to sea level rise is estimated to be more than \$3,000 billion (Nicholls et
27 634 al., 2007). A past study by Larsen et al. (2008) estimated about \$3.6–\$6.1 billion future
28 635 costs to the Alaska public infrastructure from today until 2030 and 5.6–\$7.6 billion from
29 636 today until 2080. The Stern reports prepared by the British government projected that
30 637 climate change could cost the world’s economy nearly 5% of global gross domestic
31 638 product, if nations do not take action to mitigate the effect of GHGs and adapt to
32 639 projected changes in precipitation and temperature (Stern, 2007; Stern et al., 2006).
33 640 Another study for London transport raised similar concerns (Arkell and Darch, 2006).
34 641 They reported that climate change will worsen the already existing risk on London
35 642 transport infrastructure if forward planning and cost-effective mitigation measures are
36 643 not adopted. Furthermore, in London alone, there are currently more than half a million
37 644 properties at risk from flooding with an estimated asset value of £80 billion (Lavery and
38 645 Donovan, 2005). Direct damage in the event of a future tidal flood is estimated to be in
39 646 excess of £30 billion.

47 647 The increased frequency and intensity of extreme weather events, such as heat waves or
48 648 intense rainfalls, in the future has the potential to cause significant *structural damage* to
49 649 buildings and the transport infrastructure assets. Such effects are likely to take place
50 650 within a short period of time (hours/days) and an example would be the potential of
51 651 river flooding arising from extreme rainfall resulting in loss of stability of bridge
52 652 foundations due to scour. Impacts from short-lived extreme weather events are likely to
53 653 be accompanied by changes in the long-term (years/decades) deterioration of structural

1 654 materials, buildings and the elements of the transport infrastructure. These long-term
2 655 effects are derived by the changes in average climatic conditions and would affect life
3 656 expectancy and maintenance costs of structures. Consequently, a structure which has
4 657 already seen some deterioration arising from chemical pathways (as described in
5 658 Section 3) will be weaker in the event of extreme weather affecting it.
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10 659 The following section provides an example case study for demonstrating the effects of
11 660 changes in average environmental conditions on the long-term deterioration rate of a
12 661 widely used structural material, carbon steel. This is followed by a qualitative overview
13 662 of potential damages and failures expected in the built infrastructure during extreme
14 663 environmental conditions and weather events.
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18 664 **4.1. Case study: effect of changes in average environmental conditions** 19 665 **on deterioration of carbon steel**

20 666 A widely used material, carbon steel, is considered here to investigate the effect
21 667 of changing environmental conditions on thickness loss over the period between 2010
22 668 and 2090 for four different scenarios (Fig. 4). The chosen scenarios represent varying
23 669 climate conditions. The DRF proposed by Kucera et al. (2007) is used to estimate the
24 670 total material mass loss which is then converted into thickness loss by dividing the
25 671 former by the density of carbon steel (assumed as 7.86 g cm^{-3}). The key inputs to the
26 672 DRF include climate parameters and the ambient concentrations of SO_2 , PM_{10} and H^+ .
27 673 This particular DRF was chosen because of its capability to capture gradual changes in
28 674 climate parameters over time. The values for climate parameters are obtained from the
29 675 UK Climate Projections database (UKCP09, 2009) for the area of west London for the
30 676 medium emissions scenario (see Table 6). The SO_2 and PM_{10} concentrations are taken
31 677 from Brimblecombe and Grossi (2008). H^+ concentrations are not considered for the
32 678 purposes of this case study since only the trends, and not the actual absolute values, are
33 679 of interest. Scenario 1 (base case) assumes that both the climate parameters and
34 680 pollutants concentrations vary over time according to the trends shown in Table 6.
35 681 Scenario 2 (no climate parameters and pollutants change) assumes that the climate
36 682 parameters and the SO_2 and PM_{10} concentrations remain constant, at their 2010 values,
37 683 over time. Scenario 3 (no pollutants change) assumes that the climate parameters vary
38 684 as shown in Table 6 but the SO_2 and PM_{10} concentrations remain at their 2010 values
39 685 over time; this effectively aims to capture solely the effects of changing environmental
40 686 conditions. Lastly, Scenario 4 (precipitation increase) is similar to the Scenario 1 (base
41 687 case) but assumes a 20% gradual increase in precipitation up to 2090.
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54 688 A general observation from Fig. 4 is that any potential impacts of changing
55 689 environmental conditions on the amount of material loss start becoming evident after
56 690 the second half of this century. Comparison of the “base case” with Scenario 2 shows
57 691 that the effect of a potential reduction in pollutants concentrations will be a reduction in
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1 692 the thickness loss experienced by carbon steel. This effect, however, is relatively small.
2 693 For instance, a 30% decrease in the SO₂ concentration results in only a 3% reduction in
3 694 total thickness loss by the 2090s. The effects of temperature and relative humidity can
4 695 be observed by comparing Scenario 3 with Scenario 2 where in the latter it is assumed
5 696 that temperature and humidity remain at their 2010 values. Scenario 3 can be seen to
6 697 result in 3% lower thickness loss which demonstrates that the projected increase in
7 698 average temperature and reduction in average humidity is expected to have a slightly
8 699 positive effect on the deterioration rate of carbon steel. The effect of precipitation can
9 700 be seen from Scenario 4 which shows that a 20% potential increase in average annual
10 701 precipitation by 2090 over the 2010 levels will result in an 18% increase in total
11 702 thickness loss.

12 703 It should be noted that the above observations have been obtained for a specific location
13 704 in the UK (west London) and for specific SO₂ and PM₁₀ concentrations. It would be
14 705 unwise to generalise the results since these are likely to be site-specific depending on
15 706 the climatic parameters and pollutant concentrations. Nevertheless, the results in Fig. 4
16 707 have demonstrated that changing average environmental conditions may have a slightly
17 708 positive effect on the future deterioration rate of carbon steel. Thickness loss was found
18 709 to be more susceptible to changes in average precipitation amounts rather than
19 710 temperature and relative humidity values. These results agree with the findings of
20 711 detailed investigations carried out by Kallias and Imam (2012) on the reliability of
21 712 carbon steel bridge structures.

22 713 **4.2. Buildings**

23 714 Worldwide floods are currently the second costliest weather-related catastrophe
24 715 after windstorms which are likely to increase due to rising sea levels and tidal surges
25 716 (ABI, 2005). The damage caused by flooding normally affects buildings and their
26 717 contents at basement and ground level depending on the level of water reaching to them.
27 718 The risk is present for both global flooding brought by increased river flows and local
28 719 flooding due to insufficient drainage capacity under more intense rainstorms. Hall et al.
29 720 (2006) reported that the expected annual damage in England and Wales due to coastal
30 721 flooding is predicted to increase from the current £0.5 billion to £1–£13.5 billion by
31 722 2080, depending on the scenarios of climate and socio-economical change. Their earlier
32 723 study also showed that the frequency of flooding is projected to increase more on the
33 724 coasts than on rivers (Hall et al., 2005). An indication of the overall vulnerability from
34 725 flood risks in the UK is indicated by a large number (about 5 million) of people living in
35 726 approximately 1.8 million houses in UK flood plains (Stansfield, 2001). This has led to
36 727 the preparation of strategic coastal management plans, such as retreat of the vulnerable
37 728 population, as an attempt to reduce the overall risk (Vega-Leinert de la and Nicholls,
38 729 2008). Likewise, studies investigating the impacts of climate change in the United

1 730 States report that the impact of river flooding on the Boston metropolitan area has
2 731 doubled the overall cost of flood damage (Kirshen et al., 2004; Suarez et al., 2005).
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5 732 The potential occurrence of *higher* intensity as well as more frequent *gales and winds* is
6 733 likely to increase the repair costs and the risk of roof failures, chimneys and external
7 734 cladding in buildings (Sanders and Phillipson, 2003). Roof structures and the buildings
8 735 underneath are also expected to be affected by extreme precipitation events resulting, in
9 736 many cases, from inadequate drainage. There is also a risk that hotter and drier summers
10 737 and increased exposure to UV radiation due to reduction in cloud amount could lead to
11 738 possible increase in degradation of roof membranes and exterior finishes resulting in
12 739 damage (Nijland et al., 2009; UKCIP, 2003).
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18 740 Subsidence of buildings and performance of foundations due to *drying out of clay soils*
19 741 is expected to worsen (Ross et al., 2007; Sanders and Phillipson, 2003). The expected
20 742 higher temperatures and drier weather during summer will result in a decrease in soil
21 743 moisture content and therefore lead to shrinkage and subsequent potential ground
22 744 movements and settlements. These movements may affect the global stability of
23 745 buildings and may result in structural damage, especially in the cases of shallow
24 746 foundations, but it may also give rise to non-structural damage such as cracking to
25 747 walls and finishes. For example, the heat wave observed in the UK in 2003 resulted in
26 748 insurance claims for building subsidence damage over £300 million, which was nearly
27 749 twice the value of the 1995 heat wave (Hunt et al., 2006). It is estimated that this may
28 750 further increase by 50 to 100% in the future due to the changing environmental
29 751 conditions (Stansfield, 2001). The effects of vegetation and trees around buildings can
30 752 also become detrimental and cause the ground to shrink and subside because they
31 753 extract water from the soil (Page, 1998). This can be exacerbated with the effect of
32 754 warmer summers where trees will tend to extract more water resulting in higher
33 755 subsidence.
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43 756 The increase in *temperature* extremes is also expected to affect the thermal expansion
44 757 and contraction of materials leading to more intense thermal cycle movements which
45 758 can result in cracking problems associated with buildings, cladding, sealants and roofing
46 759 membranes (Graves and Phillipson, 2000; Holper et al., 2007; UKCIP, 2003). Increases
47 760 in very hot days and heat waves will also pose limits on periods of construction activity
48 761 due to health and safety concerns.
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53 762 Changes in the intensity of driving *rain* (intense rainfall combined with strong winds) in
54 763 the future may cause problems to current designs of external cladding and cavity walls
55 764 and the interface of windows and doors with walls (UKCIP, 2003). Contrary to what
56 765 might be expected considering the increase in global average temperature, freeze-thaw
57 766 damage may not decrease. It has a potential to increase because materials may be wetter
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1 767 at the onset of frost, due to the expected higher precipitation amounts leading to more
2 768 frost damage (Nijland et al., 2009).

4.3. Transport infrastructure

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6 770 Transport infrastructure systems, which are designed to be operational over a long
7 771 time period, are increasingly likely to experience the impact of climate change over
8 772 their lifetime. Reliable estimates of future climatic conditions are essential in order to
9 773 aid infrastructure owners manage the impact of climate change on both existing and
10 774 planned infrastructure. Climate change impact planning for new items of infrastructure
11 775 will ensure continuous functionality throughout their life. Assessment of the existing
12 776 infrastructure by considering climate change effects will help infrastructure managers to
13 777 plan maintenance, modification or, in extreme cases, replacement schemes.

14 778 Due to the inherent uncertainty associated with climate change predictions and
15 779 assessment of future economic costs, the challenge is to attempt quantifying the effects
16 780 of climate change. The impacts can be direct, which is related to effects on built assets
17 781 such as bridges, highways, railway lines, ports, airport, earthworks, or indirect, which
18 782 can be related, for example, to the reduction or loss of functionality of the transportation
19 783 network, costs associated with climate change adaptations, environmental and societal
20 784 economic losses.

4.3.1. Roads

21 785
22 786 The deterioration of road surface materials such as asphalt and concrete
23 787 pavements is expected to increase due to the effects of prolonged high temperatures
24 788 resulting in softening, material breakdown, cracking and loss of road surface integrity
25 789 (DoT, 2005; Holper et al., 2007; Hudson, 2004; Karl et al., 2009; TRB, 2008; Walters,
26 790 2009; Willway et al., 2008). A recent study by Anyala et al. (2011) reports that the
27 791 cumulative rut depth expected in road pavements may increase as much as three times
28 792 by the 2050s from the present day due to the effects of increased temperature owing to
29 793 climate change. The higher risk of flooding in the future will increase the cases of
30 794 subsidence and flood damage to roads. For example, the extreme summer temperatures
31 795 observed in the UK in 2003 resulted in severe road subsidence problems reaching a total
32 796 cost of nearly £40 million (Hunt et al., 2006). A case study by Hudson (2004) has
33 797 shown that annual road maintenance costs in the UK may increase up to 60 times by the
34 798 2080s due to climate change. A similar study in Australia has shown that agency costs
35 799 for pavement maintenance and rehabilitation may increase on average by 30% in the
36 800 future due to the effects of climate change (Austroads, 2004). Coastal highways will be
37 801 at increased risk from potential sea level rises and coastal flooding (Karl et al., 2009;
38 802 Koetse and Rietveld, 2009; Mills and Audrey, 2002). Flooding of road structures may
39 803 cause significant delays to the transport network due to their unavailability. For
40 804 example, Kirshen et al. (2004) estimated an 80% increase in traveller delays due to

1 805 increased incidence of flooding (both river and coastal) in the Boston area resulting in
2 806 large economic losses. The length of roads in the UK that have significant possibility of
3 807 river or tidal flooding can increase by 40% over the present value by the 2080s (CCRA,
4 808 2012). Drainage capacity requirements to highways will increase due to increased and
5 809 more frequent extreme precipitation events, storms and weathering of drains from heat
6 810 and additional vegetation (DoT, 2005; Holper et al., 2007; West and Gawith, 2005).
7 811 More intense rainfall will also increase subsidence and heave problems on highways
8 812 (Walters, 2009). Moreover, the effects of increased tree and plant growth due to future
9 813 differences in seasonal climate variations may have major consequences for road
10 814 networks (DoT, 2005; UKCIP, 2003). Traffic signs, gantries and lightning columns are
11 815 at risk from increased wind speeds and more frequent storms (Karl et al., 2009). The
12 816 latter can also result in more debris being carried out on roads, interrupting travel.

19 817 **4.3.2. Railways**

20 818 The increase in the frequency of occurrence of extreme summer temperatures is
21 819 expected to cause deformation of rail tracks and increase the risk of rail buckling,
22 820 leading to speed restrictions and causing disruption to the transport system (Baker et al.,
23 821 2010; Dobney et al., 2009; Holper et al., 2007; Karl et al., 2009; Mills and Audrey,
24 822 2002; RSSB, 2003; TRB, 2008). For example, the heat wave observed in the UK in
25 823 2003 has resulted in a large amount of buckled rails and significant delays to the rail
26 824 network which resulted in a total cost exceeding £3.5 million (Arkell and Darch, 2006;
27 825 Hunt et al., 2006). A recent risk assessment has shown that the number of rail track
28 826 buckles in the UK may increase up to four times by the 2080s, from its present value,
29 827 due to the expected increase in hot weather during summers (CCRA, 2012). The
30 828 associated cost with the rail buckles and delays caused is set to increase by 30 to 50%
31 829 by the 2080s (Dobney et al., 2010). The risk of disruptions to railway lines due to
32 830 flooding, wind damage, and landslips is also expected to increase. A case study for
33 831 Scotland, for instance, has shown that the cost impacts of climate change may increase
34 832 by as much as 40% for the rail network in the future (Metroeconomica, 2004). The
35 833 length of railway line at significant risk of flooding in the UK may increase by as much
36 834 as 35% over its present value by the 2080s (CCRA, 2012). Furthermore, sections of the
37 835 railway network that are built along coasts will be more vulnerable to coastal flooding
38 836 and storms and to sea level rises (Baker et al., 2010; Karl et al., 2009; RSSB, 2003;
39 837 West and Gawith, 2005). Increases in vegetation growth and changes in leaf fall
40 838 patterns may cause more slipperiness in rails and may have the potential of adversely
41 839 affecting railway lines and the operation of rail networks (DoT, 2005; RSSB, 2003;
42 840 UKCIP, 2003).

55 841 **4.3.3. Bridges**

56 842 Consideration of environmental conditions for the design and maintenance of
57 843 bridge structures is vital for the safety of bridge infrastructure. Bridges are most

1 844 vulnerable to natural hazards such as flooding, storms, hurricanes and winds. Statistics
2 845 on bridge collapses worldwide reveal that natural hazards are the predominant cause of
3 846 failure (Imhof, 2004). This demonstrates, bearing in mind the adverse climate change
4 847 impacts, the high risk present in the future for bridge structures and transport networks
5 848 with respect to weather-related extreme events.
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9 849 One of the key effects of climate change on the bridge population will be the increased
10 850 risk of scour of bridge piers and abutments (DoT, 2005; RSSB, 2003; TRB, 2008). This
11 851 will arise from more frequent and more intense river flooding due to the expected
12 852 increases in precipitation in the future. Scour is caused by the erosive action of flowing
13 853 water, removing sediment from around bridge foundations. This has been one of the
14 854 most common causes of bridge collapses in the past as per the failure statistics reported
15 855 by Wardhana and Hadipriono (2003), Imhof (2004), JBA Consulting (2004), and Imam
16 856 and Chryssanthopoulos (2012). Bridge collapses in the previous years attributed to
17 857 scour and erosion are evidence of the increased risk (Fleming, 2009; Sweeney, 2009). A
18 858 recent risk assessment has shown that bridge scour may increase by between 5 and 50%
19 859 over the present value by the 2080s in the UK, depending on the local bridge site
20 860 conditions (CCRA, 2012).
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28 861 From the materials point of view, bridge material durability is also expected to be
29 862 affected by climate change. Increasing temperatures and rainfall is expected to increase
30 863 the corrosion of steel in some areas (Holper et al., 2007), increase the carbonisation and
31 864 corrosion of concrete (DoT, 2005; Peng and Stewart, 2008; Yoon et al., 2007) and
32 865 accelerate their deterioration process (UKCIP, 2003). Higher summer temperatures will
33 866 need to be handled by bridge structures and this may affect the movements required at
34 867 bearings and expansion joints due to higher levels of thermal expansion affecting bridge
35 868 operations and adding to maintenance costs (Holper et al., 2007; Karl et al., 2009;
36 869 Meyer, 2006; TRB, 2008).
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43 870 Increase in wind speeds and the occurrence of storms will pose further risk to bridge
44 871 structures in terms of the stability of their decks (DoT, 2005; Karl et al., 2009).
45 872 Drainage systems for bridges may be overloaded and may need to be changed as a result
46 873 of increased precipitation and weathering from heat (DoT, 2005; Karl et al., 2009;
47 874 RSSB, 2003; Walters, 2009). Wetter winters coupled with drier summers are also
48 875 expected to have an adverse effect on soil moisture content which can lead to
49 876 foundation settlement and landslip problems (DoT, 2005; RSSB, 2003; TRB, 2008).
50 877 Potential sea level rises may affect coastal bridges, especially if they have low
51 878 clearances below the deck with storm surges and wave actions affecting the stability of
52 879 the decks (Meyer, 2006).
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58 880 **4.3.4. Seaports**

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1 881 To date, very little work has been carried out to assess the impacts of climate
2 882 change on seaports. The principal impact of climate change on ports will be in the
3 883 higher tidal levels caused by sea level rise and changes in surge and wave conditions
4 884 (ICF International, 2006; Karl et al., 2009; West and Gawith, 2005). These would lead
5 885 to increased risk of overtopping and damage to existing flood defences. Sea level rise
6 886 will also reduce the clearance under bridges in ports having an adverse impact on the
7 887 passages of ships underneath them (Gill et al., 2009). Potential increase in wind speeds
8 888 in the future can affect operations of the cranes at ports increasing the risk of accidents.
9 889 Furthermore, it was shown that climate change will result in an increase in sea water
10 890 temperature and salinity in large areas of the Atlantic and Indian Oceans (Bindoff et al.,
11 891 2007). This may lead to increased corrosion and deterioration of port structures since
12 892 this is directly affected by the concentration of chlorides in sea water (PIANC, 2008).
13 893 At a minimum, all the above impacts are likely to result in increased weather-related
14 894 delays and periodic interruption of shipping services.

22 895 **4.3.5. Airports**

23 896 Flooding of airport runways due to extreme rainfalls and sea level rises can be
24 897 considered as one of the most important risks faced by airports due to climate change
25 898 (Karl et al., 2009; Pejovic et al., 2009). The capacity of storm water collection and
26 899 drainage systems may need to be increased to accommodate the expected future
27 900 increases in rainfall. On the other hand, rising temperatures will affect the deterioration
28 901 of runway pavements in much the same way they affect highways, resulting in heat
29 902 buckling problems (Karl et al., 2009). Coastal airports will become more vulnerable
30 903 from coastal flooding and the increase in sea level rises (Peterson et al., 2006).

36 904 **4.3.6. Earthworks and embankments**

37 905 Earthworks make up a large proportion of transport networks. In the UK, there
38 906 are about 20,000 km of embankments and cuttings (Loveridge et al., 2010). In fact,
39 907 about one fourth (£20 billion) of the total asset value of major highway infrastructure in
40 908 the UK is earthworks (Walsh et al., 2007). The long-term performance and stability of
41 909 earthworks and slopes is significantly influenced by both precipitation and temperature
42 910 conditions by affecting the seasonal pore pressure distribution within them. In a
43 911 changing climate, it is possible that failure mechanisms for these structures will alter as
44 912 a result of changes in the moisture content of the soil material of the slopes. Therefore,
45 913 the risk of failure at sites where these earthwork structures are located across the
46 914 transport network will change. The increase in heavy rain events due to climate change
47 915 can be expected to have a significant effect on embankments and slopes where
48 916 landslides can take place (Karl et al., 2009; Mills and Audrey, 2002; RSSB, 2003;
49 917 Walters, 2009; West and Gawith, 2005). Clay slopes in particular will be at greater risk
50 918 in the future from increased magnitude seasonal cycles of moisture change leading to
51 919 strength degradation (Loveridge et al., 2010). As an example, the winter of 2000/1 was

1 920 the wettest on record in many parts of the UK and rainfall caused more than 100 slope
2 921 failures in southern UK alone and additional damage in Scotland (Messafer, 2008). The
3 922 incidence of landslides is projected to increase with double the number of roads in the
4 923 UK being at risk by the 2080s compared with present values (CCRA, 2012). Drier
5 924 summers will also exacerbate subsidence in earthworks which is caused by the drying
6 925 and shrinking of soil making it unstable (Baker et al., 2010; Daeid and Thain, 2002;
7 926 West and Gawith, 2005).

12 927 **4.3.7. Tunnels**

13 928 The expected increase in precipitation levels will have an adverse effect on soil
14 929 moisture levels which may put additional risk to the structural integrity of tunnels
15 930 (RSSB, 2003). Clay shrinkage is one particular threat to tunnel structures resulting in
16 931 subsidence and heave problems. Underground tunnels will be susceptible to more
17 932 frequent flooding having an adverse effect for the operation of metro systems (Karl et
18 933 al., 2009; TRB, 2008). For example, it is stated that the projected sea level increase in
19 934 the New York area by the 2080s poses the risk of covering a large number of metro
20 935 tunnel entrances (TRB, 2008). On the other hand, changes in levels of the water table
21 936 could also impact groundwater tunnels due to increased water pressure on their walls.
22 937 Such groundwater pressures increases acting on the tunnel walls can cause tunnels to
23 938 float or crack resulting in tunnel flooding.

30 939 **4.3.8. Coastal defences**

31 940 In many cases, coastal defences can be regarded as part of the transport
32 941 infrastructure since they act to protect highway and railway networks. Security breach
33 942 of coastal defences is likely to occur more frequently due to sea level rises, more intense
34 943 storm, flooding, wave loading and erosion (Hall et al., 2006; West and Gawith, 2005).
35 944 For instance, a case study carried out in the UK showed that the expected sea level rise
36 945 and larger wave heights can increase the annual probability of failure of sea walls by a
37 946 factor of 100 (Hawkes et al., 2003). A similar study has shown that if present-day
38 947 coastal defences in the UK remain unchanged, overtopping rates can increase as much
39 948 as 80% compared to the present rates (Sutherland and Gouldby, 2003).

46 949 **5. Summary, conclusions and future research challenges**

47 950 This article discussed the footprints of air pollutants and changing environment on
48 951 the sustainability of built infrastructure. The air pollutants such as SO₂, O₃ and NO_x are
49 952 corrosive gases that deteriorate building materials through chemical routes (see Tables 2
50 953 and 3). Whereas the alterations in CO₂ concentrations play a key role in changing
51 954 climate parameters through climate change effects besides directly affecting the
52 955 concrete structures through a carbonation process. The acidic nature of corrosive air
53 956 pollutants and their ambient concentrations are equally important for geologically
54 957 sensitive industrial areas where a peculiar combination of corrosive air pollutants and

1 958 climate parameters (e.g. high relative humidity) can deteriorate structures of high
2 959 importance (e.g. museums) from both the inside and outside. Tropospheric O₃ is the
3 960 powerful oxidiser and is currently of most concern because of its global background
4 961 concentrations being over the tolerance limits of materials, and not being any positive
5 962 signs of decrease in near future (Jacob and Winner, 2009). Conversely, impact of SO₂
6 963 on various building materials is expected to fall considering the forecasted decrease in
7 964 its concentration, particularly in developed countries like the USA, Germany and UK
8 965 (Ramanathan and Feng, 2009). In line with the projected trend of Grossi et al. (2008),
9 966 our first case study carried out on four materials (carbon steel, limestone, zinc and
10 967 copper) suggested a similar decreasing trend for material deterioration in London, along
11 968 with a reasonable agreement between the recession rates computed using different
12 969 DRFs with identical input parameters. It is concluded that unlike metals and stone,
13 970 extremely limited information is available on the impact of chemical pathways on
14 971 wooden, brick and concrete structures and further detailed studies are essential for
15 972 quantifying their damage.

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24 973 Literature suggests that changing environmental conditions are expected to affect the
25 974 integrity of buildings and transport infrastructure by affecting the durability of materials
26 975 employed as discussed in Section 3. The results of our second case study revealed that
27 976 projected changes in temperature and/or relative humidity will have only a modest
28 977 effect on the deterioration rate of carbon steel. The effect of changes in precipitation and
29 978 the SO₂ concentration were found to have a more significant impact on thickness loss of
30 979 carbon steel. For combating the climate change effects on built infrastructure, a
31 980 progressive control of air pollution from local-to-regional-to-global scales is required.
32 981 With the development of knowledge data base on various pollutants, risk-effect based
33 982 approaches for air pollution control have also evolved recently (Longhurst et al., 2009).
34 983 These generally concern public health and major environmental impacts associated with
35 984 the climate change. The built infrastructure also perceives advantages of tight air
36 985 pollution control measures; for example, concentrations of acidic air pollutants are
37 986 expected to decrease in many parts of the world (Brimblecombe and Grossi, 2007).
38 987 However, impacts of air pollution on various built infrastructures of historic or public
39 988 relevance are generally overlooked when appending or designing new air pollutant
40 989 regulations. One of the major reasons for this includes lack of adequate quantitative
41 990 information on these aspects. It is therefore important to develop local air quality action
42 991 plans considering the periodic evaluation of the selected built infrastructures of interest.
43 992 These inputs could then feed into to regional or global policies. Unfortunately, assessing
44 993 the impacts of air pollutants and climate change on built infrastructure is a long term
45 994 process. Therefore, long-term monitoring studies with a periodic evaluation of air
46 995 pollution and structural deterioration could help mapping the risks of corrosion damage
47 996 more precisely.

1 997 One of the most critical steps in *adapting* infrastructures is the integration of adaptation
2 998 and mitigation considerations into standards and decision-making. This will require
3 999 cost-benefit analysis, modification of technical standards and criteria to better match
4 1000 estimates of future climatic conditions and protection as well as retrofitting of existing
5 1001 infrastructure assets. Construction techniques and materials employed will require
6 1002 adjustments to better reflect the demands of potentially more variable and extreme
7 1003 climatic conditions. The whole process will require the identification and prioritisation
8 1004 of critical infrastructure assets which require immediate attention and reinforcement
9 1005 (RAEng, 2011). Although generic risk-based frameworks for assessing different
10 1006 adaptation options have been developed during the past years, there is very little
11 1007 information available on how adaptation costs compare to the potential damages of not
12 1008 adapting and how the adaptation costs would change if there were more mitigation
13 1009 actions (EEA, 2007). More work is clearly needed in this challenging area.

14 1010 A lot more focused research studies are needed for the accurate quantification of the
15 1011 deterioration of steel, concrete, brick and wood structures in the UK and other regions.
16 1012 The outputs of these studies would help the local governments to design future
17 1013 adaptation and mitigation plans for controlling air pollution and protecting the corrosion
18 1014 of the existing built infrastructure. A holistic approach, which interlinks material
19 1015 deterioration with the air pollutants and changing environmental conditions, would be
20 1016 of great value for assessing the safety performance, reliability and robustness of
21 1017 infrastructural assets. There is also a need for developing more generic, robust, and
22 1018 theoretical DRFs that can be used for mapping corrosion damage in an area, costing
23 1019 associated risk due to this damage, and designing relevant mitigation strategies. The
24 1020 corrosion data from a greater number of field studies conducted on various types of
25 1021 building materials exposed in the range of varying environmental conditions and at
26 1022 different geographical locations could help the performance evaluation of the theoretical
27 1023 DRFs.

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31 1027 **7. References**

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1479 **List of Figure captions**

1480 **Fig. 1.** Integrated impacts and interactions of air pollutants and climate parameters on
1481 built infrastructure.

1482 **Fig. 2** (a) Total CO₂-equivalent emissions of different anthropogenic gases in 2004;
1483 CO₂ include emissions from fossil fuel use (56.6%), deformation and decay of biomass
1484 (17.3%) and other sources (2.8%) whereas F-gases include fluorocarbons such as
1485 HFCs, PFCs and SF₆, and (b) CO₂-equivalent share of total anthropogenic GHGs
1486 emissions in 2004 from different sectors (IPCC, 2007); CO₂-equivalent concentration is
1487 defined as the concentration of CO₂ that would cause the same amount of radiative
1488 forcing as a given mixture of CO₂ and other forcing components.

1489 **Fig. 3.** Estimates of annual surface recession and material loss in London during 1990
1490 and 2010 for (a) limestone, (b) carbon steel, (c) zinc, and (d) copper.

1491 **Fig. 4.** Effect of changing environmental conditions on thickness loss of carbon steel.

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1493 **List of Tables**

1494 **Table 1.** Corrosion sensitivity of various materials towards air pollutants and climate
 1495 parameters. The list is compiled based on the various DRFs proposed in the literature
 1496 (Kucera and Fitz, 1995; Kucera et al., 2007; Leuenberger-Minger et al., 2002; Noah's
 1497 Ark, 2006; Scheffer, 1971; Screpanti and De Marco, 2009; Tidblad et al., 2001).

Material	SO ₂	O ₃	NO ₂	PM ₁₀	Rh	Rain[H ⁺]	Rain [Cl ⁻]	Temp
Weathering steel	⊗	⊗		⊗	⊗			⊗
Carbon steel	⊗	⊗		⊗	⊗			⊗
Limestone	⊗	⊗	⊗	⊗	⊗	⊗		⊗
Sandstone	⊗				⊗	⊗		⊗
Zinc	⊗	⊗			⊗	⊗		⊗
Aluminium	⊗				⊗		⊗	⊗
Copper	⊗	⊗			⊗	⊗		⊗
Cast Bronze	⊗	⊗				⊗	⊗	⊗
Glass	⊗		⊗					⊗
Paint/galvanised	⊗							⊗
Paint/steel	⊗							⊗
Nickel	⊗	⊗						⊗
Tin		⊗						⊗
Rubber and plastic materials		⊗						⊗
Lead					⊗			⊗
Wood						⊗	⊗	⊗
Concrete	⊗		⊗		⊗	⊗	⊗	⊗

Table 2. GHGs and their global warming potential (Forster et al., 2007; IPCC, 2007; Ravishankara et al., 1993; UNEP, 2007).

GHGs	Global mean Concentrations in 2005 (ppm)	Atmospheric life time (~years)	Change since 1998 (ppm)	Radiative forcing (W m ⁻²)	GWP for 100-year time with respect to CO ₂
CO ₂	379 ± 0.65	500	+13	1.66	1
CH ₄	1.774 ± 0.0018	12	+0.011	0.48	25
N ₂ O	0.319 ± 0.00012	114	+0.005	0.16	298
CFCs ^b	868 ± 0.604	45-640	-13	0.257	4,750–6130
SF ₆ ^b	5.6 ± 0.038	up to 3200	+1.5	0.0029	22,800

^aAccepted value range though this is variable and can be estimated by the formulae given in foot note ‘a’ on page 213 of the IPPC (2007) report.

^bCFCs include combined net values (in ppt) of CFC–11, CFC–12 and CFC–113. SF₆ values are in ppt.

Table 3. List of DRFs for various materials in unsheltered conditions. Both Tidblad et al. (2001) and Screpanti and De Marco (2009) gives DRFs for 8–year exposure. DRFs given by Leuenberger–Minger et al. (2002) and Noah's Ark (2006) are for 4 and 1 year exposure, respectively. ML, LL and R stand for mass loss by corrosion attack in g m^{-2} , leached layer in nm, surface recession or thickness loss in μm (>1–year exposure) or $\mu\text{m yr}^{-1}$ (1–year exposure), respectively. Gaseous and ion concentrations are annual mean in $\mu\text{g m}^{-3}$ and mg lit^{-1} . D_{cl} is chloride deposition ($\text{mg m}^{-2} \text{day}^{-1}$) and $\text{Rh}_{60} = (\text{Rh} - 60)$ when $\text{Rh} > 60$; otherwise 0. Rn is precipitation in m yr^{-1} ; V_{dS} and V_{dN} are deposition velocities (cm s^{-1}) for SO_2 and HNO_3 , respectively.

Material	Dose–response function	Source
Weathering steel	$\text{ML} = 34[\text{SO}_2]^{0.33} e^{[0.02\text{Rh} + f_{\text{ws}}(\text{T})]} t^{0.33}$ $f_{\text{ws}}(\text{T}) = 0.059(\text{T} - 10)$ when $\text{T} \leq 10^\circ\text{C}$ $f_{\text{ws}}(\text{T}) = -0.036(\text{T} - 10)$ when $\text{T} > 10^\circ\text{C}$	Tidblad et al. (2001)
	${}^a\text{R} = 1.92 + 2.97[\text{SO}_2] t_{\text{ow}} t^{0.37} + 0.89[\text{SO}_2] t_{\text{ow}} u t^{0.37} + 0.15[\text{O}_3] t^{0.37}$	Leuenberger–Minger et al. (2002)
Carbon steel	$\text{R} = 1.58[\text{SO}_2]^{0.52} e^{[0.02\text{Rh} + f_{\text{Cs}}(\text{T})]} + 0.166\text{Rn}[\text{H}^+] + 0.0761 \text{PM}_{10} + 0.102\text{D}_{\text{Cl}}^{0.33} e^{[0.033\text{Rh} + 0.040\text{T}]}$ $f_{\text{Cs}}(\text{T}) = 0.150(\text{T} - 10)$ when $\text{T} \leq 10^\circ\text{C}$ $f_{\text{Cs}}(\text{T}) = -0.054(\text{T} - 10)$ when $\text{T} > 10^\circ\text{C}$	Noah's Ark (2006)
	$\text{R} = 1.77[\text{SO}_2]^{0.52} e^{[0.20\text{Rh} + f_{\text{ws}}(\text{T})]} + g(\text{Cl}^-, \text{Rh}, \text{T})$ $\text{ML} = 29.1 + t^{0.6} (21.7 + 1.39[\text{SO}_2]^{0.6} \text{Rh}_{60} e^{f_{\text{ws}}(\text{T})} + 1.29 \text{Rn}[\text{H}^+] + 0.593\text{PM}_{10})$ $\text{F}_{\text{Cs}}(\text{T}) = 0.150(\text{T} - 10)$ when $\text{T} \leq 10^\circ\text{C}$ $\text{F}_{\text{Cs}}(\text{T}) = -0.054(\text{T} - 10)$ when $\text{T} > 10^\circ\text{C}$ $g(\text{Cl}^-, \text{Rh}, \text{T})$ is a function describing the dry deposition effect of chloride in combination with Rh and T	Kucera et al. (2007)
	$R = 13.4t^{0.98} \left(\frac{t_{\text{ow}}}{3800} \right)^{0.46} \left(1 + \frac{[\text{SO}_2]}{25} \right)^{0.62} \left(1 + \frac{[\text{D}_{\text{cl}}]}{50} \right)^{0.34} e^{0.016(\text{T} + 20)}$	Klinesmith et al. (2007)
Steel panels with alkyde (fresh samples)	$\text{ML} = (0.033 [\text{SO}_2] + 0.013\text{Rh} + f_{\text{Sp}}(\text{T}) + 0.0013\text{Rn}[\text{H}^+]) t^{0.41}$ $f_{\text{Sp}}(\text{T}) = 0.015(\text{T} - 11)$ when $\text{T} \leq 11^\circ\text{C}$ $f_{\text{Sp}}(\text{T}) = -0.15(\text{T} - 11)$ when $\text{T} > 11^\circ\text{C}$	Tidblad et al. (2001)

1 2 3 4 5 6	Coil coated galvanised steel with alkyd melamine (fresh samples)	$ML = (0.0084[SO_2] + 0.015Rh + f_{Cc}(T) + 0.00082 Rn[H^+]) t^{0.43}$ $f_{Cc}(T) = 0.040(T-10)$ when $T \leq 10^\circ C$ $f_{Cc}(T) = -0.064(T-10)$ when $T > 10^\circ C$	Tidblad et al. (2001)
7 8 9 10 11 12 13 14 15 16 17 18	Portland limestone	$R = 2.7 [SO_2]^{0.48} e^{-0.018T} t^{0.96} + 0.019 Rn[H^+] t^{0.96}$ $R = 3.1 + t(0.85 + 0.0059 Rh_{60} [SO_2] + 0.078 Rh_{60} [HNO_3] + 0.054Rn[H^+] + 0.0258 PM_{10})$ $[HNO_3] = 516 e^{-3400/(T+273)} ([NO_2][O_3] Rh)^{0.5}$	Tidblad et al. (2001) Kucera et al. (2007) Lipfert (1989)
19 20 21 22 23		$R = 18.8 Rn + 0.016 [H^+] Rn + 0.18 (V_{ds} [SO_2] + V_{dN} [HNO_3])$ ${}^cR = 3.1 + t (0.85 + 0.0059[SO_2] Rh_{60} + 0.054 Rn[H^+] + 0.078 (516 e^{-3400/(T+273)} ([NO_2] [O_3] Rh)^{0.5} Rh_{60}) + 0.0258 PM_{10})$	Screpanti and De Marco (2009)
24 25 26 27 28	White Mansfield sandstone	$R = (2[SO_2]^{0.52} e^{f_{Ms}(T)} + 0.028Rain[H^+])t^{0.91}$ $f_{Ms}(T) = 0$ when $T \leq 10^\circ C$ $f_{Ms}(T) = -0.013(T-10)$ when $T > 10^\circ C$	Tidblad et al. (2001)
29 30 31 32 33 34 35 36 37 38 39	Zinc	$ML = 1.4[SO_2]^{0.22} e^{[0.018Rh + f_{Zn}(T)]} t^{0.85} + 0.029Rn[H^+] t$ $f_{Zn}(T) = 0.062(T-10)$ when $T \leq 10^\circ C$ $f_{Zn}(T) = -0.021(T-10)$ when $T > 10^\circ C$	Tidblad et al. (2001)
40 41 42 43 44		$ML = 1.82 + t (1.71 + 0.471[SO_2]^{0.22} e^{[0.018Rh + f_{Zn}(T)]} + 0.041Rn[H^+] + 1.37[HNO_3])$ $f_{Zn}(T) = 0.062(T-10)$ when $T \leq 10^\circ C$ $f_{Zn}(T) = -0.021(T-10)$ when $T > 10^\circ C$	Kucera et al. (2007)
45 46 47 48 49 50		${}^aR = 0.33 + 0.38[SO_2] t_{ow} u t^{0.53} - 0.5[SO_2] t_{ow} [O_3] t^{0.53} + 0.00007[SO_2][O_3] Rh t^{0.53}$	Leuenberger-Minger et al. (2002)
51 52 53		$R = 0.196[SO_2]^{0.22} e^{[0.018Rh + f_{Zn}(T)]} + 0.00406 Rn[H^+] + 0.192 [HNO_3] + 0.0175D_{Cl}^{0.57} e^{[0.008Rh + 0.085T]}$ $f_{Zn}(T) = 0.062(T-10)$ when $T \leq 10^\circ C$ $f_{Zn}(T) = -0.021(T-10)$ when $T > 10^\circ C$	Noah's Ark (2006)
54 55 56 57	Aluminium	$R = 0.16t^{0.36} \left(\frac{t_{ow}}{3800} \right)^{0.24} \left(1 + \frac{[SO_2]}{25} \right)^{0.82} \left(1 + \frac{[D_{cl}]}{50} \right)^{0.44} e^{0.05(T+20)}$ $ML = 0.0021[SO_2]^{0.23} Rh e^{f_{Al}(T)} t^{1.2} + 0.000023 Rn[Cl^-] t$ $f_{Al}(T) = 0.031(T-10)$ when $T \leq 10^\circ C$ $f_{Al}(T) = -0.61(T-10)$ when $T > 10^\circ C$	Klinesmith et al. (2007) Tidblad et al. (2001)

	$R = 0.094t^{0.05} \left(\frac{t_{ow}}{3800} \right)^{0.23} \left(1 + \frac{[SO_2]}{25} \right)^{1.14} \left(1 + \frac{[D_{cl}]}{50} \right)^{0.42} e^{0.01(T+20)}$	Klinesmith et al. (2007)
Copper	$ML = 0.0027 [SO_2]^{0.32} [O_3]^{0.79} Rh e^{f_{Cu}(T)} t^{0.78} + 0.050$ $Rn[H^+] t^{0.89}$ $f_{Cu}(T) = 0.083(T-10) \text{ when } T \leq 10^\circ C$ $f_{Cu}(T) = -0.032(T-10) \text{ when } T > 10^\circ C$	Tidblad et al. (2001)
	$ML = 3.12 + t (1.09 + 0.00201 [SO_2]^{0.4} [O_3] Rh_{60} e^{f_{Cu}(T)} + 0.0878 Rn[H^+])$ $f_{Cu}(T) = 0.083(T-10) \text{ when } T \leq 10^\circ C$ $f_{Cu}(T) = -0.032(T-10) \text{ when } T > 10^\circ C$	Kucera et al. (2007)
	${}^a R = 0.1 + 0.2[SO_2] t_{ow} t^{0.41} + 0.0044[SO_2][O_3] t^{0.41} + 0.016[O_3] t^{0.41}$	Leuenberger-Minger et al. (2002)
	$R = 0.000302[SO_2]^{0.32} [O_3]^{0.79} Rh e^{f_{Cu}(T)} + 0.00560 Rn[H^+] + 0.0125D_{Cl}^{0.27} e^{[0.036Rh + 0.049T]}$ $f_{Cu}(T) = 0.083(T-10) \text{ when } T \leq 10^\circ C$ $f_{Cu}(T) = -0.032(T-10) \text{ when } T > 10^\circ C$	Noah's Ark (2006)
	$R = 0.46t^{0.15} \left(\frac{t_{ow}}{3800} \right)^{0.02} \left(1 + \frac{[SO_2]}{25} \right)^{0.38} \left(1 + \frac{[D_{cl}]}{50} \right)^{0.46} e^{0.02(T+20)}$	Klinesmith et al. (2007)
Cast Bronze	$ML = 0.026 [SO_2]^{0.44} Rh e^{f_{Br}(T)} t^{0.86} + 0.029 Rn[H^+] t^{0.76} + 0.000043 Rn[Cl^-] t^{0.76}$ $f_{Br}(T) = 0.060(T-11) \text{ when } T \leq 11^\circ C$ $f_{Br}(T) = -0.067(T-11) \text{ when } T > 11^\circ C$	Tidblad et al. (2001)
	$ML = 1.33 + t (0.00876 [SO_2] Rh_{60} e^{f_{Br}(T)} + 0.0409 Rn[H^+] + 0.038 PM_{10})$ $f_{Br}(T) = 0.060(T-11) \text{ when } T \leq 11^\circ C$ $f_{Br}(T) = -0.067(T-11) \text{ when } T > 11^\circ C$	Kucera et al. (2007)
	$R_{Br} = 0.0255R_{steel}$	Noah's Ark (2006)
Glass	$LL = 0.013 [SO_2]^{0.49} Rh^{2.8} t$	Tidblad et al. (2001)
	${}^d LL = -0.28 + 0.028Rh - 0.055 T$	Noah's Ark (2006)
Lead	${}^b ML = 0.5(0.0125D_{Cl}^{0.27} e^{(0.036Rh+0.049T)} + 0.0175D_{Cl}^{0.57} e^{(0.008Rh+0.085T)})$	Noah's Ark (2006)
Wood (Scots pine sapwood and Douglas fir heartwood)	${}^e d = d_{mc} \times d_t$ $d_{mc} = 6.75 \times 10^{-10} MC^5 - 3.5 \times 10^{-7} MC^4 + 7.18 \times 10^{-5} MC^3 - 7.22 \times 10^{-3} MC^2 + 0.34 MC - 4.98 \text{ (if } MC > 25\%)$ $d_t = 1.8 \times 10^{-6} T^4 + 9.57 \times 10^{-5} T^3 - 1.55 \times 10^{-3} T^2 + 4.17 \times 10^{-2} T$ $\text{(If } T_{min} > -1^\circ C \text{ and } T_{max} < 40^\circ C)$	Brischke and Rapp (2008)

1 1518 ^a t_{ow} is the time of wetness i.e. fraction of exposure duration with temperature >0°C and Rh>80%; u is
2 1519 the annual mean wind speed (m s⁻¹) at 10 m above the ground level.
3 1520 ^b D_{Cl} is chloride deposition in mg m⁻² d⁻¹.
4 1521 ^cGaseous and PM₁₀ concentrations are annual mean in mg m⁻³.
5 1522 ^dLL is in μm.
6 1523 ^e d is total daily dose, d_{mc} is moisture content (MC) induced daily dose and d_t is temperature induced
7 1524 daily dose; T , T_{min} and T_{max} are daily average wood temperature, minimum and maximum
8 1525 temperatures, respectively. These functions are developed up to 7 years exposure.

18 1526 **Table 4.** List of DRFs for calculation of the first-year corrosion loss of various
19 1527 structural metals, given by BS EN ISO 9223 (2012). Here, r_{corr} is the first-year
20 1528 corrosion rate of metals in μm yr⁻¹, T is the annual average temperature in °C, Rh is the
21 1529 annual average relative humidity in %, P_d is the annual average SO₂ deposition in mg
22 1530 m⁻² day⁻¹, S_d is the annual average Cl⁻ deposition in mg m⁻² day⁻¹.
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Material	DRFs
Carbon steel	$r_{corr} = 1.77P_d^{0.52} e^{[0.020Rh + f_{St}]} + 0.102S_d^{0.62} e^{[0.033Rh + 0.040T]}$ $f_{St} = 0.150(T-10) \text{ when } T \leq 10^\circ\text{C}$ $f_{St} = -0.054(T-10) \text{ when } T > 10^\circ\text{C}$
Zinc	$r_{corr} = 0.0129P_d^{0.44} e^{[0.046Rh + f_{Zn}]} + 0.0175S_d^{0.57} e^{[0.008Rh + 0.085T]}$ $f_{Zn} = 0.038(T-10) \text{ when } T \leq 10^\circ\text{C}$ $f_{Zn} = -0.071(T-10) \text{ when } T > 10^\circ\text{C}$
Copper	$r_{corr} = 0.0053P_d^{0.26} e^{[0.059Rh + f_{Cu}]} + 0.01025S_d^{0.27} e^{[0.036Rh + 0.049T]}$ $f_{Cu} = 0.126(T-10) \text{ when } T \leq 10^\circ\text{C}$ $f_{Cu} = -0.080(T-10) \text{ when } T > 10^\circ\text{C}$
Aluminium	$r_{corr} = 0.0042P_d^{0.73} e^{[0.025Rh + f_{Al}]} + 0.0018S_d^{0.60} e^{[0.020Rh + 0.094T]}$ $f_{Al} = 0.009(T-10) \text{ when } T \leq 10^\circ\text{C}$ $f_{Al} = -0.043(T-10) \text{ when } T > 10^\circ\text{C}$

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1 1534 **Table 5.** Overview of climate change impacts on the structural integrity of built
 2
 3 1535 infrastructure.

Climate change effect on built infrastructure	Climate change variables				
	Temperature	Precipitation	Humidity	Wind	Sea level rise
Building material deterioration	⊗	⊗	⊗		
Building structural integrity	⊗	⊗		⊗	⊗
Building roofs	⊗	⊗		⊗	
Building foundations	⊗	⊗			
Bridge material deterioration	⊗	⊗	⊗		
Bridge scour		⊗			⊗
Bridge thermal movements	⊗				
Bridge deck stability				⊗	⊗
Bridge foundation settlements/landslips	⊗	⊗			
Pavement deterioration	⊗	⊗	⊗		⊗
Railway lines	⊗	⊗			⊗
Highway / bridge drainage	⊗	⊗			
Traffic signs/lighting				⊗	
Seaports				⊗	⊗
Airports	⊗	⊗			⊗
Earthworks / embankments	⊗	⊗			
Tunnels		⊗			⊗
Coastal defences					⊗

42 1536 **Table 6.** Summary of climate parameters and pollutant concentrations considered for
 43
 44 1537 estimating the deterioration of carbon steel during the period 2010-2090.

Year	[SO ₂] (μg m ⁻³)	Rh (%)	T (°C)	Rn (mm)	PM ₁₀ (μg m ⁻³)
2010	17	78.9	11.2	643	30
2030	15	78.4	11.8	643	15
2050	14	77.7	12.5	645	14
2070	13	77.0	13.3	645	13
2090	12	76.6	14.0	646	12

54 1538

Figure 1

[Click here to download Figure: Fig. 1.ppt_Flow chart-Impact and interactions.ppt](#)

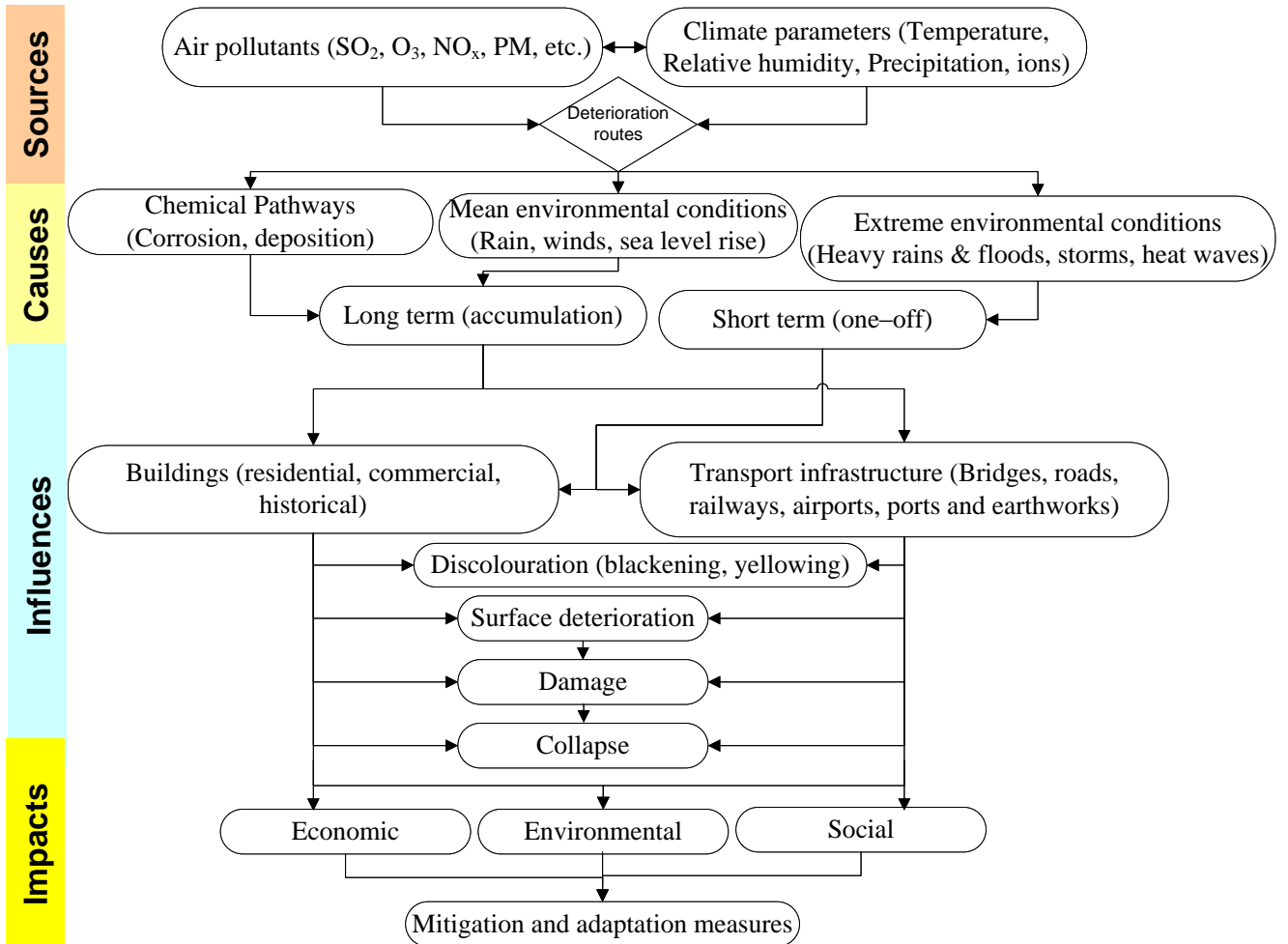


Figure 2

[Click here to download Figure: Fig. 2.ppt_CO2 equivalent emissions.ppt](#)

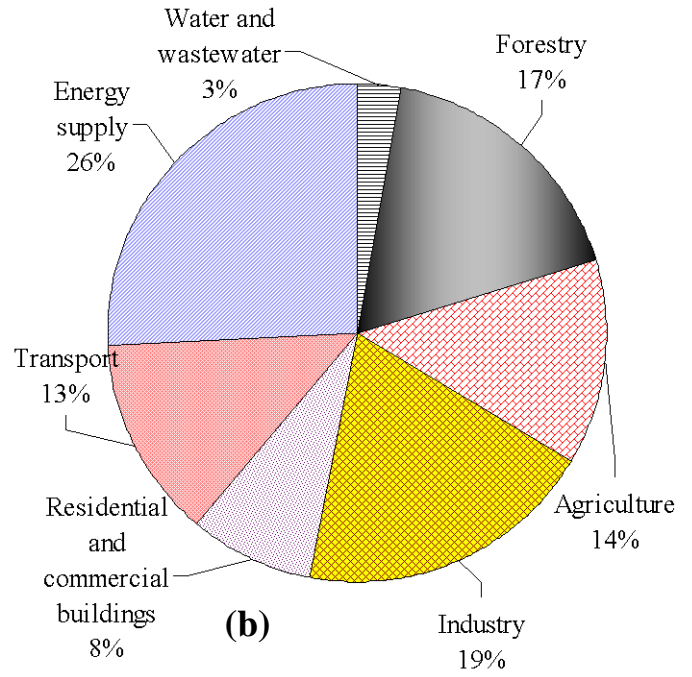
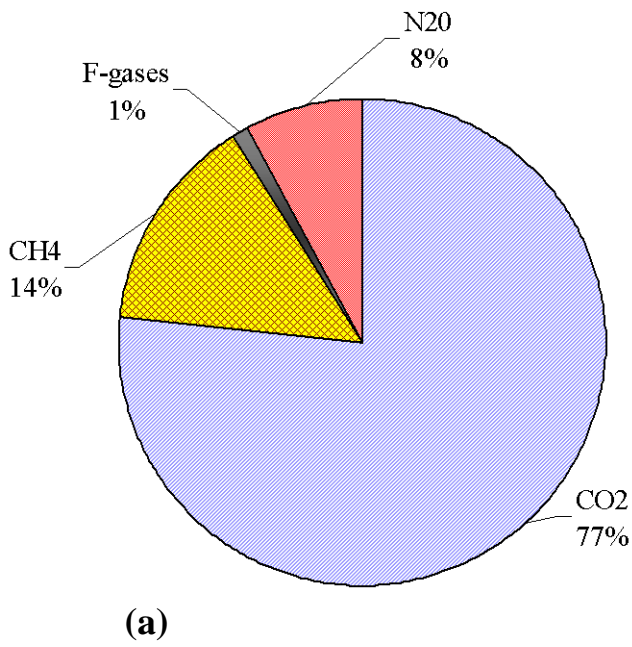


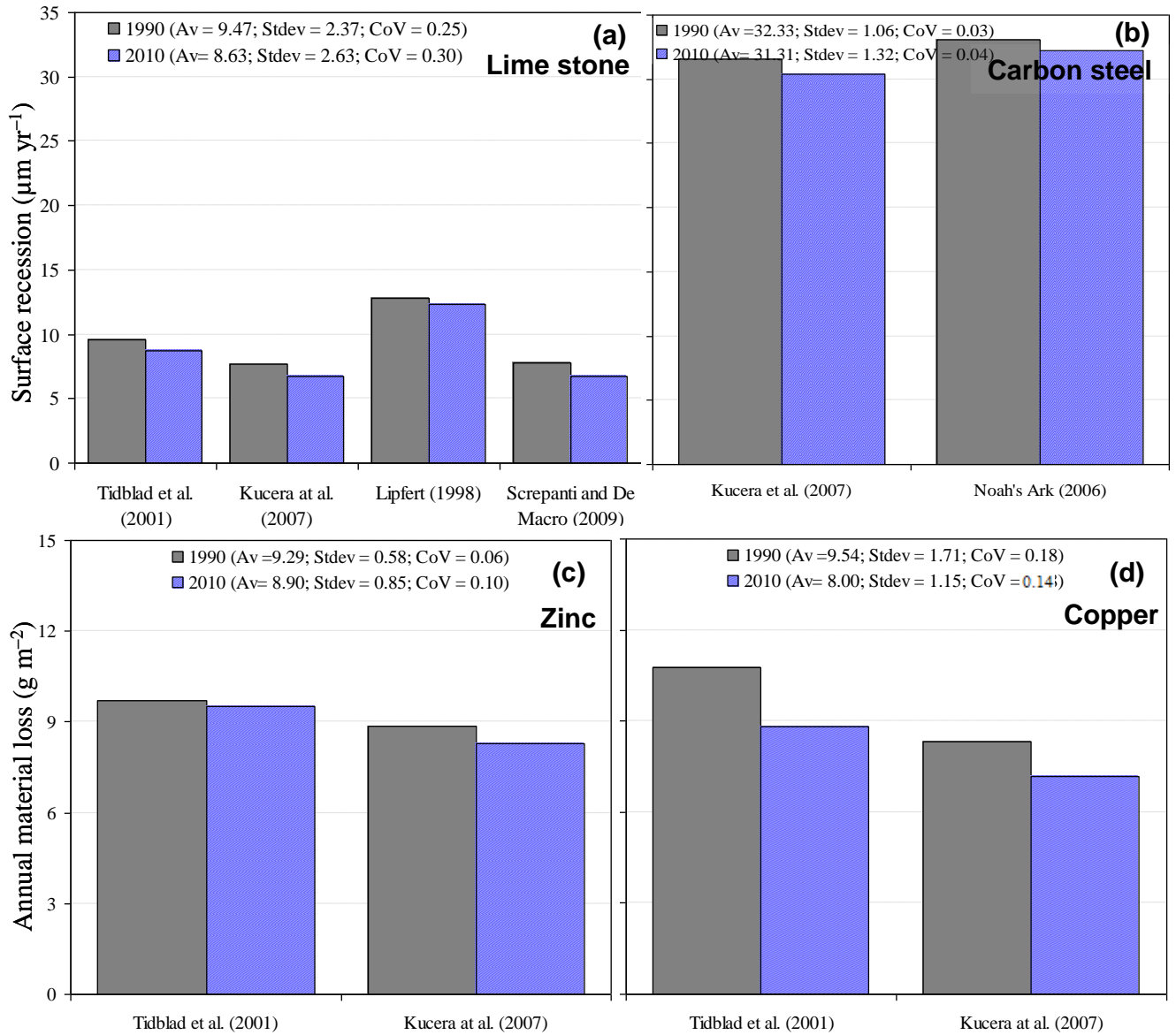
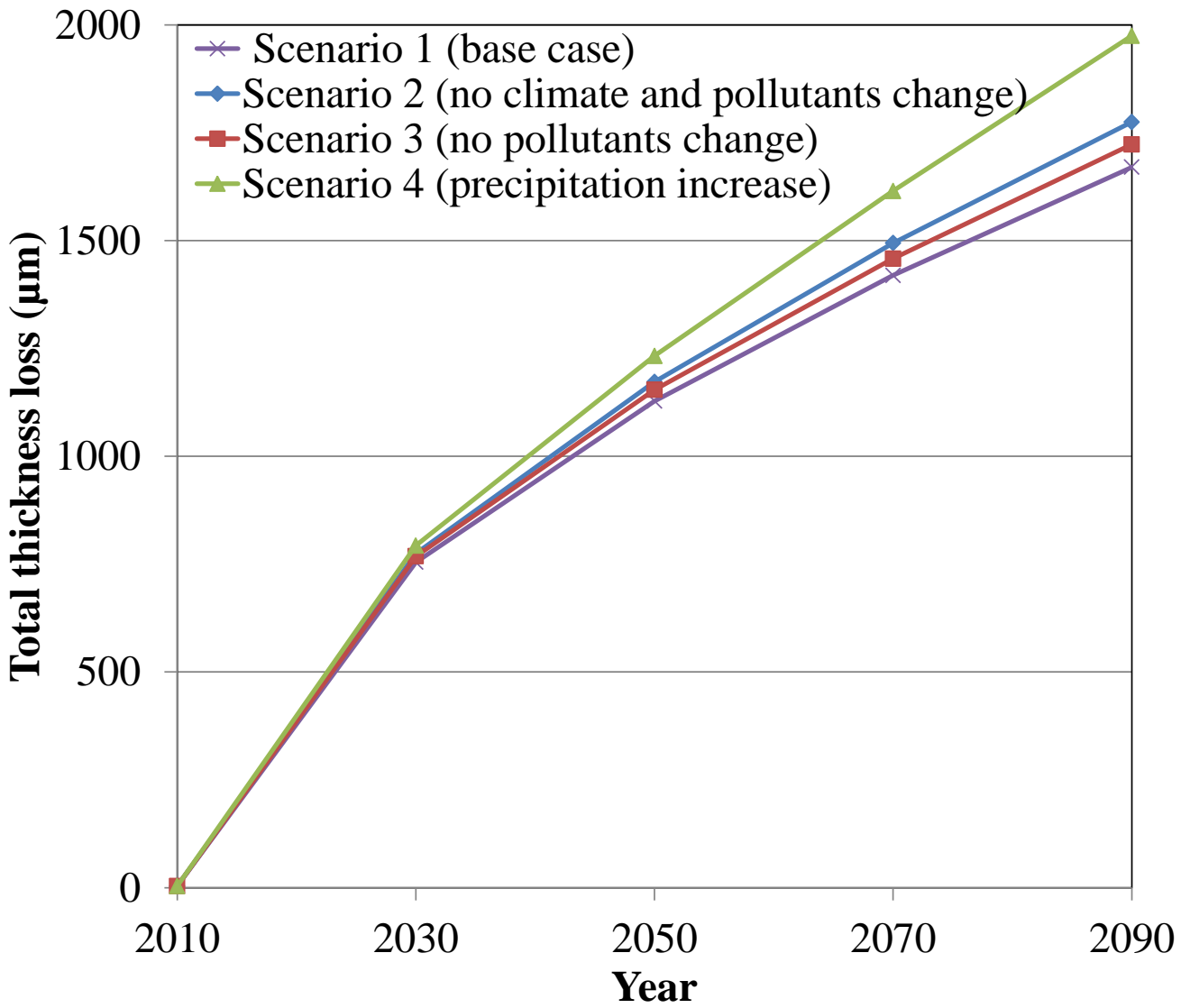
Figure 3[Click here to download Figure: Fig. 3.ppt_Surface recession in London.ppt](#)

Figure 4

[Click here to download Figure: Fig. 4.ppt_Carbon Steel_Case Study.pptx](#)



Research Highlights

- ▶ Impacts of air pollution and changing environment on built infrastructure is reviewed
- ▶ Chemical sensitivity of various building materials is assessed
- ▶ Inventory of DRFs, and their application through a case study, is carried out
- ▶ Both air pollution and changing environment affect the integrity of built structures
- ▶ Robust and generalised DRFs are needed for mapping corrosion losses accurately

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<http://dx.doi.org/10.1016/j.scitotenv.2012.11.056>